

APPLICATION OF SELECTIVE LEACHING IN FABRICATION OF THIN FILM YBCO DEVICES.

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Abstract--Ethylenediamine solutions have been shown to turn bulk YBCO into insulating materials. In this work, the effect of these solutions on thin YBCO films is studied. In both unpatterned and patterned films, a smooth decrease in critical currents, a transition to normal state and a subsequent gradual increase in resistance is observed as a function of exposure time to the solution. These characteristics might make this process desirable for weak link and on-film resistor fabrication.

I. INTRODUCTION

Some thin film device fabrication processes require controlled weakening of the superconductor [1,2]. Turning superconductor into resistive material is also of use in thin film circuit fabrication when on-film resistors or RF matched loads are required. In this work we assess the usefulness of ethylenediamine solutions to achieve both effects.

These solutions have already been shown to preferentially leach the copper from bulk YBCO samples, leaving behind materials known to be insulating [3].

When YBCO films are exposed to such solutions, leaching creates a layer whose copper concentration has been reduced from its original value. Auger profiles have shown that the relative concentration of copper in a partially leached film increases gradually from a minimum value at the surface to a constant, stoichiometric value at some depth within the film. This depth is dependent on the exposure of the film to the ethylenediamine solution. The combined effect of material modification and reduction of the thickness of the superconducting layer results in a reduction of the critical current and an increase in resistance once the sample is no longer superconductive.

A potential advantage of this process with respect to the conventional method of weak link fabrication (etching film away in the weak link regions [1,4,5]) is enhanced controllability, since as unlike for etching, diffusion tends to slow down the leaching process as deeper film layers are chemically modified.

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II. EXPERIMENTAL

A. Materials

Films from several sources were used in this work and wide variations in reactivity were observed from film to film. Most of the films were prepared by Bruce Davidson at the University of Wisconsin, Madison by laser ablation on LaAlO₃ and MgO substrates. These films produced the most repeatable results.

Olin Hunt HR-100 negative PR and Hoechst Celanese AZ1350J positive PR were used for patterning and later processing. The leaching solution was prepared from ethylenediamine (Aldrich 99% ethylenediamine -- used as shipped) and triply distilled water as a 2.0 M solution and used at room temperature. Contacts were made by sputter depositing silver, which was later annealed in oxygen (1 atm) at 375 °C for 60 min. Epo-Tek H31 silver epoxy, cured for 4 hr at 100 °C, was used to bond 36 gauge copper wires to the pads.

B. Procedures

Mechanical tests: Thickness versus leach experiments were performed on several patterned films using profilometry. Measurement error was highest in films with LaAlO₃ substrates due to the terraces formed by twinning. No appreciable change in film thickness was seen in any of the films measured, as shown in Fig. 1. This was also confirmed by SEM images.

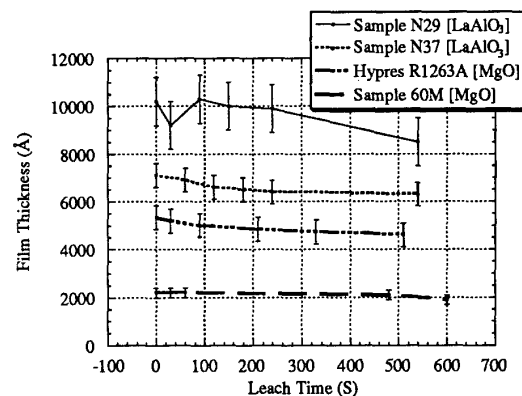


Fig. 1 Thickness vs. exposure to 2.0 M ethylenediamine solution for several different YBCO films, as measured by profilometry.

Electrical tests in unpatterned films: The critical current and critical temperature were measured in several unpatterned films using a 4 point probe setup. Figures 2 and 3 show a typical set of such results. Smooth transitions from metallic to insulating behavior were observed above the critical temperature in the transition diagrams as the leaching time was increased, and no significant changes in critical temperature were found as long as the sample remained superconductive (a resistivity drop was still detected when samples were not fully superconducting). Below T_C , the critical currents were also gradually depressed as the films were exposed to the leaching solutions. Eventually, the films became normal conductors. Once this happened, their resistivity could be gradually increased by continuing the exposure of the films to the leachant. Such increases could be carried out over many orders of magnitude.

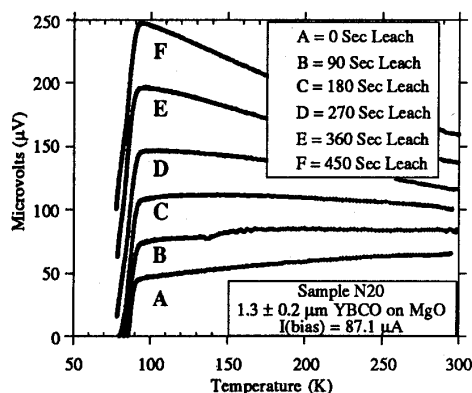


Fig. 2 Transition diagram of an unpatterned film with leaching time as a parameter.

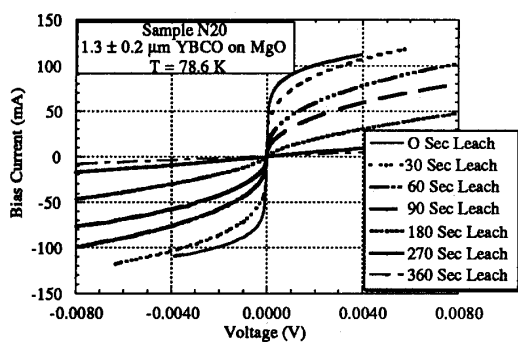


Fig. 3 I-V Curves of an unpatterned with leaching time as a parameter.

Electrical tests in patterned films: Films were patterned using AZ1350J photoresist and a HNO_3 solution [1]. Six

links of varying sizes were patterned in each film, and the films were covered with HR-100 resist with openings on the links and contact areas. This resist withstands thermal cycling, which allows leaching and testing of the samples in cyclic sequence.

Preliminary results show transition diagrams and I-V curves qualitatively comparable to Figs. 2 and 3. Comparison among links in the same film is indicated in Fig. 4.

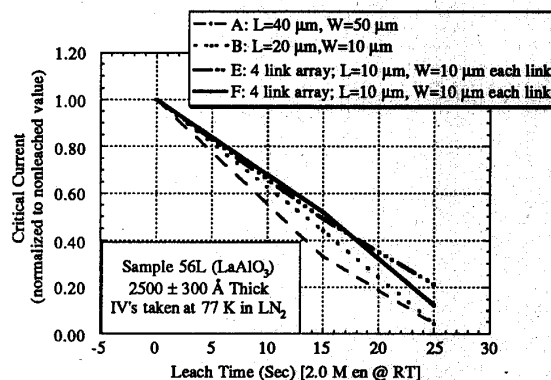


Fig. 4 Relative reduction in critical current of four links of varying lengths (L) and widths (W) patterned in the same film.

At this point it is suspected that residues from silver paint solvents are deposited in the links after each leaching stage, and they interfere with the leaching process. Such residues could come from the silver paint (used to attach the film substrate to a copper block) dissolving when the sample is rinsed with methanol after leaching. It is hoped that greater uniformity throughout the film will be obtained once this is solved.

III. CONCLUSION

Ethylenediamine solutions are being tested in thin film device applications. Results on unpatterned films indicate adequate behavior of these solutions for weak link and on-film resistor fabrication. This has been qualitatively reproduced in patterned films. Achieving uniformity in the characteristics of several links within a film still presents practical problems which are currently being addressed.

IV. ACKNOWLEDGMENT

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