

1 Liquid crystalline thermosets based on anisotropic
2 phases of cellulose nanocrystals

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12 nanocomposite, lyotropic liquid crystals.

16 **ABSTRACT**

17 A new class of liquid crystalline thermosets (LCTs) was successfully produced containing lyotropic
18 cellulose nanocrystals (CNCs) as the primary mesogenic component (up to 72 % wt.) by the addition of
19 non-mesogenic epoxy monomers. Cellulose-based LCTs were produced by totally aqueous processing
20 methods and ultimately cured at elevated temperatures to produce **ordered networks** of ‘frozen’ liquid
21 crystalline (LC) phases. Various degrees of birefringence were obtained via self-assembly of cellulose
22 nanocrystals into oriented phases as observed by polarized optical microscopy and transmission
23 electron microscopy. X-ray diffraction measurements highlighted the effects of texture of CNCs within
24 LCT films compared to lyophilized CNCs. Cellulose-based LCT films uniquely exhibited thermo-
25 mechanical properties of both traditional liquid crystalline thermosets and liquid crystalline elastomers,
26 such as high elastic modulus (~1 GPa) under ambient conditions and low glass transition temperature
27 (~ -25 °C), respectively. The development of liquid crystalline thermosets based on cellulose
28 nanocrystals and aqueous processing methods provides a renewable pathway for designing high
29 performance composites with **ordered network structures and unique optical properties.**

30

31 **Introduction**

32 “Freezing” liquid crystalline (LC) phases into a highly cross-linked and ordered polymeric
33 network can lead to desirable mechanical and optical properties (Barclay and Ober, 1993; Ortiz et al.,
34 1998a; Shiota and Ober, 1997). Traditionally, this has been accomplished by cross-linking of rigid-rod
35 molecules containing thermotropic mesogens end-capped with functional groups such as acrylates,
36 cyanates, or glycidyl groups. These highly ordered and densely cross-linked network structures
37 prepared by polyaddition/polycondensation processes resulted in a new class of materials termed liquid
38 crystalline thermosets (LCTs). Another related class of materials, called liquid crystalline elastomers
39 (LCEs), provide more loosely cross-linked networks prepared by similar processes (Giamberini et al.,
40 2005; Ortiz et al., 1998b; Ribera et al., 2006). The primary advantage of LCTs is enhanced mechanical
41 properties due to heterogeneities and localized anisotropy of nematic structure within an overall
42 isotropic microstructure (Kannan and Sudhakara, 2011). These heterogeneities are suggested to cause
43 deviations in crack propagation from a straight line leading to increased modulus and fracture
44 toughness. In addition to having microscopic domains of uniform orientation, LCTs and LCEs can also
45 be macroscopically oriented by curing in an electric or magnetic field. As one of the most widely used
46 thermosetting materials, epoxy resins represent an important class of LCTs due to unique chemical
47 characteristics such as curing reactions without byproducts, low shrinkage, and controllable cross-
48 linking (Carfagna et al., 1994, 1997). Epoxy-based LCTs can also display unique optical properties due
49 to their anisotropy, such as birefringence useful for nonlinear optics, filters and photorefractive
50 materials (Amendola et al., 1996; Tan and Fung, 2001). Epoxy resins, in general, have a broad range of
51 applications including adhesives, coatings, and composite materials (May and Tanaka, 1973). Their
52 high stiffness and low density make epoxies ideal matrices in fiber-reinforced composites, especially
53 for wind power (Brondsted et al., 2005) and the aircraft industry (Kannan and Sudhakara, 2011).

54 Cellulose nanocrystals (CNCs) afford significant improvements in the thermo-mechanical
55 properties of a wide range of polymer matrices by adding only small amounts (Habibi et al., 2010;
56 Moon et al., 2011). Also referred to as cellulose whiskers or nanocrystalline cellulose (NCC), cellulose
57 nanocrystals consist of rod-like nanoparticles with widths ranging 5-70 nm and lengths of 100 nm to
58 several microns depending on the source of biomass, thus provide a high aspect ratio reinforcing phase
59 (Bras et al., 2011). Epoxy nanocomposites have been produced using cellulose nanocrystals as a
60 reinforcing phase, while most researchers have focused on solvent exchange, gel templating or
61 lyophilizing CNCs and re-dispersion in the epoxy hardener (Rusli and Eichhorn, 2008, 2011) or in
62 polar organic solvents such as dimethyl formamide (Khelifa et al., 2012; Tang and Weder, 2010).
63 Nevertheless, Ruiz et al. (Ruiz et al., 2000; Ruiz et al., 2001) demonstrated the reinforcing effect of up
64 to 5 % wt. cellulose nanocrystals within a thermoset matrix via a water-based epoxy emulsion system,
65 which avoided solvent-exchange and lyophilization. Given the abundance and biodegradability of
66 cellulose, and the desire to decrease our dependence on petroleum-based chemicals, it would be
67 beneficial to develop high performance composites, which contain cellulose as the major component,
68 *i.e.* the polymer matrix. Some examples of epoxy-based nanocomposites containing high cellulose
69 content have been developed in the form of highly transparent and thermally conductive films using
70 bacterial cellulose (Retegi et al., 2012) and nanofibrillated cellulose (Shimazaki et al., 2007),
71 respectively. However, to the best of our knowledge, the use of cellulose nanocrystals (CNCs) as the
72 major component of liquid crystalline epoxy thermosets has been unexplored.

73 In addition to the high stiffness and thermal stability of cellulose nanocrystals, **negative surface**
74 **charges derived from sulfuric acid hydrolysis promote uniform dispersions stabilized via electrostatic**
75 **repulsions (Habibi et al., 2010). This phenomenon leads to their self-assembly into liquid crystalline**
76 **(LC) phases originally proposed to form helical chiral nematic structures** (Revol et al., 1992).

77 However, recently Picard et al. (Picard et al., 2012) have proposed a new model in which CNCs are
78 arranged into multilamellar smectic liquid crystals wherein each liquid plane contains nematic phases.
79 Thus, the precise structure of CNC liquid crystalline phases is currently a topic of debate. Such LC
80 phases can also be preserved upon air-drying to produce iridescent solid films. The LC phase of
81 cellulose nanocrystals is sensitive to processing conditions (Pan et al., 2010) and has been exploited as
82 a template to create novel solid-state materials containing periodic nanostructures with tunable optical
83 properties (MacLachlan et al., 2011; Shopsowitz et al., 2012; Shopsowitz et al., 2010). Additionally, a
84 number of patents have been filed which describe methods for incorporating patterns (Beck et al.,
85 2010b) and controlling the iridescent wavelength (Beck et al., 2010a) of cellulose nanocrystal films by
86 differential heating and ultrasound/high shear, respectively. Proposed applications of such films include
87 coatings for security and optical authentication devices.

88 A major disadvantage of cellulose nanocrystal films is their tendency to be brittle and have poor
89 flexibility, which is an important feature for practical applications in most fields. Mixing aqueous
90 dispersions of CNCs with appropriate water-soluble polymers or monomers offers a possible solution,
91 however partial loss of liquid crystalline ordering of CNCs is evident at high concentrations of
92 additives. Zou et al. (Zou et al., 2010) demonstrated flexible and iridescent cellulose nanocrystals films
93 by adding up to 25% wt. poly(vinyl alcohol) with low molecular weight. Recently, Tatsumi et al.
94 (Tatsumi et al., 2012) have developed poly(2-hydroxyethyl methacrylate) (PHEMA) nanocomposites
95 reinforced with 5 % wt. cellulose nanocrystals which have been locked in their liquid crystalline
96 assembly via *in-situ* polymerization of HEMA. Khelifa et al. (Khelifa et al., 2012) also highlighted
97 interesting optical properties of nanocomposites based on an acrylic copolymer matrix bearing epoxy
98 moieties loaded with 20% wt. CNCs, which was ascribed to the formation of LC phases. Other notable
99 works have focused on the use of external forces to produce anisotropic nanocomposites of CNCs, such

100 as a magnetic field (Pullawan et al., 2012) and compression molding (Rusli et al., 2010). Although, to
101 date, ‘freezing’ the liquid crystalline assembly of cellulose nanocrystals into free-standing films by the
102 addition of thermosetting epoxy monomers in aqueous media has not been reported.

103 In this work, we demonstrate a simple method for producing a new class of liquid crystalline
104 thermosets based on lyotropic liquid crystals, *i.e.* cellulose nanocrystals, and water-soluble non-
105 mesogenic epoxy monomers. These LCTs were produced by totally aqueous processing methods and
106 ultimately cured at elevated temperatures to produce **ordered networks** of ‘frozen’ LC phases. Films
107 were produced by casting cellulose nanocrystal/monomer mixtures from aqueous dispersions at
108 ambient conditions and subsequent curing at elevated temperatures. X-ray diffraction experiments
109 highlighted the effects of texture of CNCs within LCT films as compared to lyophilized CNCs. These
110 cellulose-based LCTs exhibited thermo-mechanical properties which uniquely combined those of
111 traditional LCTs and LCEs. All of the processed LCT films exhibited various degrees of birefringence
112 as observed by polarized optical microscopy. In addition, the LCT films selectively attenuated the
113 entire spectrum of UV wavelengths. Microscopic and highly localized domains of oriented cellulose
114 nanocrystals were confirmed by transmission electron microscopy. The development of liquid
115 crystalline thermosets based on cellulose nanocrystals and aqueous processing methods is expected to
116 open renewable pathways for designing high performance composites with **ordered network structures**
117 **and unique optical properties.**

118

119 **Materials and Methods**

120 *Materials*

121 Sulfuric acid (95%) and acetone (99%) were purchased from VWR Scientific (Helsinki, Finland).
122 Dialysis tubing cellulose membrane (MWCO 12,400), 1,4-butanediol diglycidyl ether (BDGE), 2,2'-
123 (ethylenedioxy)bis(ethylamine) (EBEA) were all purchased from Sigma-Aldrich (Helsinki, Finland).

124

125 *Preparation of cellulose nanocrystals (CNCs)*

126 CNCs were obtained from cotton fibers by acid hydrolysis with 65 wt% aqueous sulfuric acid solution
127 (45 °C, 45 minutes). The resulting dispersion was diluted and filtered into ~200 g ice cubes and
128 washed with distilled water until neutral pH by successive centrifugations at 12,000 rpm at 4 °C for 20
129 minutes. Finally, dialysis for one week against distilled water with a 12,400 MWCO dialysis
130 membrane was performed to remove trace amounts of residual sulfuric acid. Concentrations of the
131 resulting CNCs dispersions were calculated gravimetrically to be 3.11% wt. CNCs utilized to make
132 LCT films were in the acid form with a solution pH = 5.9.

133

134 *Film casting*

135 Aqueous dispersions of cellulose nanocrystals (3.11% wt.) were combined with stoichiometric
136 mixtures of 1,4-Butanediol diglycidyl ether (BDGE) (epoxy resin) and 2,2'-
137 (Ethylenedioxy)bis(ethylamine) (EBEA) (diamine hardener). For simplicity, the molar ratio of epoxy
138 resin to diamine hardener was kept at 1:1. The three components were stirred for 30 min at room
139 temperature and then poured into polystyrene Petri dishes (50 mm). The films were casted by slow
140 solvent evaporation under ambient conditions for four days. At this point, the films were initially cured
141 at 60 °C for 24 hours, carefully removed from the polystyrene dishes, and finally cured at 150 °C in a
142 vacuum oven for 1 hour. The resulting LCT films were calculated to contain 50-72 % wt. (or ~42-66 %
143 v/v) CNCs. A neat epoxy film was also prepared under the same conditions for comparison.

144

145 *Atomic Force Microscopy (AFM)*

146 CNC dispersions were diluted to the desired concentration and spin-coated onto UV-treated silicon
147 wafers with an anchoring polymer at 4000 rpm. The deposition of the suspension was performed on a
148 substrate that was then spun for 30 seconds. The surfaces were then dried at 80 °C for 1 hour. The
149 spin-coated CNC films were imaged using a scanning probe microscope (model Multimode Nanoscope
150 IIIa, Digital Instruments) in tapping mode, which was equipped with an E or J scanner and with non-
151 coated silicon-etched probes (model LTESP, Veeco) with a nominal spring constant of 48 N/m.

152

153 *Attenuated Total Reflectance-Fourier Transform Infrared Spectroscopy (ATR-FTIR)*

154 Infrared spectra were obtained in order to confirm the polymerization of epoxy monomers in the
155 presence of CNCs. Dried samples were directly analyzed in a Mattson 3000 FTIR spectrometer
156 equipped with a Pike Technologies GladiATR accessory. All spectra were collected with a 2 cm⁻¹
157 resolution after 32 continuous scans.

158

159 *Differential Scanning Calorimetry*

160 Differential scanning calorimetry (DSC) was performed using a Mettler Toledo Star DSC 821e. In a
161 typical experiment *ca.* 10 mg of dry sample was placed in the DSC cell and first heated from -50 to 200
162 °C, followed by a second ramp from -50 to 225 °C (heating rate of 10 °C min⁻¹). The glass transition
163 temperature T_g and change in heat capacity ΔC_p were calculated from the second heating cycle using
164 Mettler Toledo STARe software.

165

166 *Dynamic Mechanical Analysis*

167 DMA measurements were performed using a DMA Q800 with rectangular films (*ca.* 10 x 5.5 x 0.35
168 mm) in tensile mode with an oscillation frequency of 1 Hz, a static force of 10 mN, and an oscillation
169 amplitude of 5 μm . Measurements were carried out at a heating rate of 3 $^{\circ}\text{C min}^{-1}$ in the temperature
170 range of -100 to 150 $^{\circ}\text{C}$. Tensile tests of films were also carried out at 35 $^{\circ}\text{C}$ under 3 N/min isothermal
171 ramp. All measurements were performed at ambient humidity. Rectangular sample sections were taken
172 from the centers of each film for consistency between measurements.

173

174 *UV/Vis-spectrophotometer*

175 Ultraviolet-visible spectroscopy was conducted on a UNICAM HELIOS β UV/Vis spectrophotometer.
176 Transmission spectra of the samples in the wavelength range of 190 nm to 1000 nm were collected by
177 mounting free-standing LCT films so that the surfaces of the films were perpendicular to the beam
178 path.

179

180 *Polarized optical microscopy*

181 Optical birefringence of LCT films was visualized with an Olympus BH-2 microscope equipped with a
182 full-wave retardation plate and Nikon Coolpix 990 digital camera. All images were taken with the
183 polarizer and analyzer in a crossed (orthogonal) arrangement at low magnification, *i.e.* zoom lens 5X
184 and 10X. All photographs were taken with the same illumination conditions and exposure time.

185

186 *X-ray Diffraction (XRD)*

187 XRD experiments were carried out at ambient temperature on freeze-dried CNCs, a pure CNC film and
188 LCT films in powder mode. Samples of freeze-dried CNCs were prepared by powder placement in the
189 sample holder and manually pressing with a flat plate, while flexible LCT films were simply placed in

190 **the sample holder.** A PANalytical X'Pert PRO MPD diffractometer operated with a CuK_α anode (λ =
191 0.154 nm) in parallel geometry was used with a 2θ range from 5° to 40° with steps of 0.0263° and a
192 scanning speed of 2.67°/min. The crystallinity index (*CrI*) was calculated by the method of Segal et al.
193 (Segal et al., 1959) according to:

$$CrI = 1 - \frac{I_{am}}{I_{200}}$$

194
195 where I_{200} is the maximum intensity (arbitrary units) of the (200) lattice diffraction and I_{am} is the
196 intensity of diffraction (in the same units) at $2\theta = 18^\circ$.

197 The dimension of the crystal D_{hkl} was evaluated from a Lorentzian peak function of the (200) lattice
198 diffraction by using Scherrer's formula (Klug and Alexander, 1954):

$$D_{hkl} = \frac{0.9 \times \lambda}{\beta_{1/2} \times \cos \theta}$$

199
200 where θ is the diffraction angle, λ the X-ray wavelength and $\beta_{1/2}$ the peak width at half of maximum
201 intensity.

202

203 **Transmission Electron Microscopy**

204 LCT films were sectioned using a microtome and deposited on copper mesh grids and observed with a
205 JEOL JEM-3200FSC transmission electron microscope (TEM) operated at an accelerating voltage of
206 200 kV.

207

208 **Results and Discussion**

209 *Preparation of cellulose nanocrystals and liquid crystalline thermosets*

210 To date, most efforts to produce epoxy nanocomposites containing cellulose nanocrystals have
211 been focused on their use as matrix reinforcement up to *ca.* 24 % (v/v) (Ruiz et al., 2000; Ruiz et al.,
212 2001; Rusli and Eichhorn, 2008, 2011; Tang and Weder, 2010). Contrarily, the objective of this work
213 was to produce liquid crystalline thermosetting (LCT) films using cellulose nanocrystals as the primary
214 mesogenic component by the addition of water-soluble non-mesogenic epoxy precursors. To this
215 effect, cellulose nanocrystals were produced from cotton fiber by sulfuric acid hydrolysis to form
216 highly stable aqueous dispersions. The resulting CNCs were determined to have widths ranging ~4-12
217 nm and lengths from ~100-200 nm according to AFM height images (see Fig. S1 in Supplementary
218 data), which were comparable to recent literature (Habibi et al., 2010).

219 It is widely known that CNCs spontaneously self-assemble into liquid crystalline phases at a
220 specific range of concentrations, which depends on their aspect ratio (Habibi et al., 2010; Moon et al.,
221 2011; Picard et al., 2012; Revol et al., 1992) as dictated by the Onsager theory of colloidal liquid
222 crystals (Onsager, 1949). The LC phase is preserved upon slow evaporation of water to produce
223 iridescent films with tunable optical properties (Picard et al., 2012; Revol et al., 1998). However, CNC
224 films are typically brittle which provides a challenge for their use in practical applications. In an effort
225 to produce CNC films which preserve their LC assembly and introduce flexibility, a water-soluble
226 epoxy monomer, 1,4-butanediol diglycidyl ether (BDGE), and a diamine hardener, 2,2'-
227 (ethylenedioxy)bis(ethylamine) (EBEA) were mixed with aqueous dispersions of CNCs. The mixtures
228 were film casted in polystyrene dishes by slow solvent evaporation over about four days. Once the
229 films were dry, they were partially cured at 60 °C for 24 hours, carefully removed from the polystyrene
230 dishes, followed by final curing at 150 °C for 1 hour. The resulting LCT films contained between 50-72
231 % wt. CNCs, as calculated gravimetrically from aqueous dispersions (3.11% wt.). An attempt was
232 made to produce LCT films containing higher compositions of CNCs, however they were too brittle to

233 be removed from the Petri dishes without rupturing. A neat epoxy film was also prepared by mixing the
234 epoxy monomer with diamine hardener and casting, following the same curing schedule.

235 To confirm the curing reaction between the epoxy monomer and diamine hardener in the
236 presence of CNCs, the LCT films were characterized by ATR-FTIR spectroscopy. Shown in Figure 1
237 are FTIR spectra of lyophilized CNCs, neat epoxy film, and the highest (72 % wt.) and lowest (50 %
238 wt.) CNC compositions of LCT films. All other LCT films displayed similar spectra and for simplicity
239 are not shown here. As compared to lyophilized CNCs, typically the LCT films showed changes in
240 peak intensities in the range from 1500-1000 cm^{-1} , which corresponded to various C-O bonds present
241 in the cured epoxy. Characteristic peaks corresponding to an epoxide ring (950-900 cm^{-1}) were slightly
242 overlapped by the cellulose spectrum such that its presence could not be accurately detected.
243 Additionally, the most obvious peak appeared at 1650 cm^{-1} and corresponded to a secondary amine (N-
244 H) bending vibration of the cured epoxy in the LCT film. The same characteristic peak at 1650 cm^{-1}
245 was also observed in a neat cured epoxy film, which confirmed success of the curing reaction in the
246 presence of CNCs.

247

248 *Optical properties*

249 The resulting LCT films were easily removed and flexible, however they were not iridescent as
250 compared to pure CNC films by the naked eye. Also, slight darkening of LCT films was evident which
251 was attributed to the presence of acidic sulfate groups on CNCs surfaces as shown in the small inset of
252 Figure 2. The films were 300-400 μm thick, and noted to be slightly thicker near the center. Since the
253 LCT films were casted from water in polystyrene dishes, the variations in film thickness were likely
254 due to a complex synergy between surface energy phenomena associated with water droplet
255 evaporation from hydrophobic surfaces (Picknett and Bexon, 1977), the “coffee ring effect” (Uetani

256 and Yano, 2012) and the intermolecular interactions of epoxy monomers with CNCs. This may have
257 caused heterogeneous distributions of epoxy monomers and CNCs within LCT films, thus variations in
258 film thickness. When viewed under crossed-polarizers, the entirety of all the LCT films displayed some
259 degree of birefringence, but brilliant color patterns spanning the visible spectrum were observed
260 predominantly near the film edges, as shown in Figure 2a. On the other hand, away from the film
261 edges, variations such as brighter (Figure 2b) and darker areas (Figure 2c) were observed under
262 crossed-polarizers. A neat epoxy film showed no birefringence as indicated by complete lack of light
263 transmittance when viewed between crossed-polarizers (not shown here). This was an indication that
264 the cured epoxy resin **alone** was isotropic, such that any birefringence observed in LCT films was
265 **derived from orientations caused by the presence of CNCs.**

266 The birefringence observed near LCT film edges was similar to that of locally oriented
267 anisotropic glass as previously described for CNCs cast under vacuum (Viet et al., 2007). However,
268 variations in film thickness may have resulted from heterogeneous distributions of CNCs and cured
269 epoxy within the LCT films, thus significant differences in color patterns and overall birefringence was
270 observed. For example, the darker textures, such as in Figure 2c, indicated optically isotropic areas
271 such that low concentrations of oriented CNCs and high concentrations of cured epoxy were present.
272 These findings indicated that **opportunities might exist to fine tune the optical properties** of cellulose-
273 based LCT films by varying the processing conditions (Beck et al., 2010a, b). **The observed**
274 **birefringence** would suggest that CNCs were able to self-assemble into oriented phases more
275 effectively at the film edges, possibly due to lower concentrations of non-mesogenic epoxy monomers
276 during solvent evaporation. **However, the fingerprint texture characteristic of CNCs periodic liquid**
277 **crystalline structure was not observed.** A pure CNC film cast under similar conditions, shown in Figure
278 2d, displayed typical color textures arising from their periodic LC structures (Picard et al., 2012; Revol

279 et al., 1998), but these characteristics were not observed for any of the LCT films. Evidently, the
280 presence of non-mesogenic epoxy monomers disrupted the formation of periodic LC phases during the
281 solvent evaporation process. Overall, the observed birefringence of LCT films, although
282 heterogeneous, gave indirect evidence for CNCs assembled into various domains of oriented phases,
283 which was later confirmed by transmission electron microscopy.

284 The 'frozen' LC phases present in iridescent CNC films is known to cause specific attenuations
285 in the visible region of the electromagnetic spectrum, which has been associated with the helical pitch
286 of chiral nematic structures (Shopsowitz et al., 2010). However, Picard et al. (Picard et al., 2012) would
287 argue that such attenuations are associated with lamella thickness of smectic phases and explained by
288 classical light interference. UV-Vis spectra of a pure CNC film, neat epoxy film, and the highest (72 %
289 wt.) and lowest (50 % wt.) CNC compositions of LCT films are shown in Figure 3. All other LCT
290 films displayed similar UV spectra and for simplicity are not shown here. In this case, pure CNC films
291 gave an attenuation peak at ~600 nm. In contrast, the neat epoxy film and LCT films did not show a
292 peak, which suggested a lack of periodic LC structure. The neat epoxy film was semi-transparent (~60-
293 70 % transmittance) in the visible region, while LCT films were only transparent at higher wavelength
294 ranges approaching near infrared. Nevertheless, the presence of CNCs in the LCT films yielded a ~100
295 nm redshift, which led to selective attenuation of the entire UV wavelength range. This was most likely
296 due to the slight darkening of LCT films as shown in the inset of Figure 2. Nevertheless, these results
297 suggested that LCT films could also be useful as functional composites that require UV attenuation,
298 having similar properties to cellulose composites that incorporated TiO₂ nanoparticles as UV absorbers
299 (Zhou et al., 2006).

300

301 *Cellulose degree of crystallinity*

302 X-ray diffraction measurements were carried out to determine cellulose degree of crystallinity
303 and reveal the effects of texture of CNCs within LCT films. Shown in Figure 4 are the XRD spectra of
304 the four LCT films produced containing 50-72 % wt. CNCs, a neat epoxy film, a pure CNC film and a
305 sample of lyophilized CNCs. As expected, the neat epoxy film did not show any diffraction peaks in
306 the $2\theta = 5-40^\circ$ range, which indicated a completely amorphous material. Samples of lyophilized CNCs
307 showed the typical XRD spectrum for cellulose I (Nishiyama et al., 2002; Park et al., 2010), displaying
308 2θ diffraction peaks at 14.8 (1-10), 16.5 (110), 22.8 (200) and 34.5° (004). The same diffraction peaks
309 were observed in all of the LCT films, although notable differences in peak intensities were observed,
310 especially in peaks from (1-10) and (110) reflections. In the case of lyophilized CNCs, which were
311 thought to be preferentially oriented along the nanocrystal axis by mechanical pressing (Klug and
312 Alexander, 1954), the peak intensities for (1-10) and (110) reflections were relatively low compared to
313 (200). However, a notable increase in intensity for (1-10) relative to the (110) and (200) reflections was
314 observed for all the LCT films, presented in Table 1 as an intensity ratio (I_{200}/I_{1-10}). This suggested that
315 while lyophilized CNCs existed with a preferred axial orientation (along the fiber axis) (Klug and
316 Alexander, 1954; Yoshiharu et al., 1997), this effect progressively decreased for CNCs in LCT films as
317 indicated by more intense (1-10) reflections relative to (200) in Figure 4. A pure and iridescent CNC
318 film was also analyzed to confirm the differences in peak intensities due to preferred orientation (top of
319 Figure 4). The pure CNC film, which is known to contain periodic LC phases (Picard et al., 2012) (see
320 Figure 2d), displayed a similar increase in the (1-10) reflection relative to (200). Even though pure
321 CNC films were highly ordered into periodic phases, their domains were too small to be detected by
322 XRD (Yoshiharu et al., 1997). It has been proposed that CNC periodic phases contain differently
323 shaped bundles of CNCs within different lamella (Picard et al., 2012), and the precise arrangement of
324 CNCs within each bundle is unknown. On the other hand, the traditional chiral nematic model of CNCs

325 (Revol et al., 1992) provides precise helicoidal arrangements. The results of XRD indicate here that the
326 overall microstructure of pure CNC films appeared more isotropic (randomly oriented) as compared to
327 lyophilized CNCs. This suggested that CNCs in LCT films were oriented along many different
328 directors, which could partially explain the overall apparent lack of texture compared to lyophilized
329 CNCs. While these findings were not taken as evidence of periodic LC phases in LCT films, the
330 significant changes in relative peak intensities compared to lyophilized CNCs highlighted the
331 importance of texture on routine interpretations of XRD spectra of cellulose (Park et al., 2010).

332 From the XRD spectra, the intensity ratio (I_{200}/I_{1-10}), crystallinity indices and average crystallite
333 widths were calculated based on Scherrer's formula with a shape constant of 0.9 (Klug and Alexander,
334 1954), and are summarized in Table 1. Lyophilized cotton CNCs gave a crystallinity index of 90.7 %
335 and crystallite width of 6.6 nm, which was in good agreement with AFM images (see Fig. S1 in
336 Supplementary data). The crystallinity index and crystallite width of CNCs within the LCT films
337 generally increased with increasing amount of CNCs. Interestingly, LCT films gave much higher
338 crystallinity indices than would have been expected based on the film compositions. For example, LCT
339 films containing only 50% wt. CNCs gave a crystallinity index of 85.1 %, which was only ~5 % less
340 than pure lyophilized CNCs. In addition, the average crystallite width showed slight increases as the
341 CNC composition of LCT films was increased, up to 7.0 nm, however this was considered to be within
342 experimental error given the textural effects observed in Figure 4. The pure and iridescent CNC film
343 displayed a crystallinity index of 85.6 %, which was surprisingly lower than lyophilized CNCs. Indeed,
344 the Scherrer method assumes that crystallites are randomly oriented (Klug and Alexander, 1954), and
345 preferred orientation of crystallites is known to affect peak intensity and broadening. Thus, calculations
346 of crystallinity index and crystallite dimensions were believed to be affected by differences in overall
347 orientation of cellulose crystallites in the films. Also, the method of Segal et al. (Segal et al., 1959) to

348 calculate the crystallinity index (CrI) from XRD spectra has been known to overestimate cellulose
349 **degree of crystallinity** as compared to other methods, such as solid-state ^{13}C -NMR (Park et al., 2010).
350 It is important to note that the calculations of crystallinity index (CrI) from XRD spectra only provided
351 an empirical measure of relative cellulose **degree of crystallinity**.

352 The intensity ratio of the (200) and (1-10) reflections, (I_{200}/I_{1-10}), was calculated from XRD
353 spectra as a qualitative comparison of textural effects between CNC samples. For lyophilized CNCs,
354 the ratio (I_{200}/I_{1-10}) was the highest and assumed to be due to preferred axial orientation of crystallites
355 by mechanical pressing (Klug and Alexander, 1954). Interestingly, the intensity ratio (I_{200}/I_{1-10}) for
356 LCT films displayed a trend based on the film compositions, such that as the concentration of CNCs
357 was increased, the intensity ratio (I_{200}/I_{1-10}) decreased. This indicated that the overall preferred axial
358 orientation of CNCs became less pronounced at higher concentrations. The pure and iridescent CNC
359 film showed the lowest ratio (I_{200}/I_{1-10}), which correlated well with the trend observed for LCT films.
360 As previously noted, the relative peak intensities were not taken as an indication of periodic LC
361 structures, however the variations in peak intensities between samples highlighted the importance of
362 preferred orientation on routine interpretations of XRD spectra of cellulose (Park et al., 2010).

363

364 *Thermo-mechanical properties*

365 Besides the unique optical properties exhibited by epoxy-based liquid crystalline thermosets
366 (Amendola et al., 1996; Tan and Fung, 2001), desirable thermo-mechanical properties **such as**
367 **increased modulus and fracture toughness** are expected to result from their highly ordered networks
368 (Barclay and Ober, 1993; Carfagna et al., 1997; Kannan and Sudhakara, 2011; Ortiz et al., 1998a).
369 Thermal transitions of LCT films were initially observed by DSC and the results are summarized in
370 Table 2. Samples of lyophilized CNCs, LCT films and a neat epoxy film were first subjected to a

371 temperature ramp from -50 to 200 °C, followed by a second cycle from -50 to 225 °C. The first
372 temperature cycle was only brought to 200 °C so to avoid any thermal degradation of cellulose. After
373 removal of thermal history, lyophilized CNCs showed a glass transition (T_g) of 211 °C, in good
374 agreement with other highly crystalline forms of cellulose I (Szczesniak et al., 2008). Contrarily, it was
375 clear that the only thermal transition that could be observed in LCT films was the glass transition (T_g)
376 of the cured epoxy resin. The glass transition of cellulose is highly dependent on degree of crystallinity,
377 moisture content and thermal history, such that the T_g of CNCs within LCT films could not be
378 observed. The pure amorphous epoxy film showed a T_g of -21.9 °C, while the T_g 's of LCT films were
379 insignificantly decreased by ~1-2 °C. Due to the low epoxy content of the LCT film containing 72 %
380 wt. CNCs, no thermal transitions were observed. From the glass transitions, changes in heat capacity,
381 ΔC_p , were also determined from each sample and correlated to the film compositions. As shown in
382 Table 2, the experimentally determined compositions of the LCT films followed a reasonable trend in
383 correlation with the stoichiometric values, based on initial content of aqueous dispersions. However,
384 the inherent heterogeneities and variations in LCT film thickness may have caused some deviation in
385 the experimentally determined compositions.

386 Typically, pure CNC films are very brittle and challenging for dynamic mechanical analysis
387 (Bras et al., 2011), thus CNC stiffness and deformation have been determined within epoxy
388 nanocomposites by techniques such as Raman spectroscopy (Rusli and Eichhorn, 2008, 2011; Sturcova
389 et al., 2005). In this case, we obtained flexible LCT films containing up to 72 % wt. CNCs which were
390 highly durable for mechanical analysis. Rectangular sample sections were taken from the center of each
391 LCT film for consistent analysis between samples. The evolution of storage modulus (E') of LCT films
392 and neat epoxy was monitored over a temperature range of -100 to 150 °C by Dynamic Mechanical
393 Analysis (DMA), shown in Figure 5(a). As compared to the neat epoxy film, all of the LCT films

394 exhibited significantly higher moduli over the entire temperature range. This was expected since the
395 LCT films contained cotton cellulose nanocrystals as the major component, which have a reported
396 theoretical modulus between 57-105 GPa (Rusli and Eichhorn, 2008). Additionally, all the LCT films
397 were mechanically stable up to the onset of thermal degradation of CNCs at ~ 225 °C (data not shown
398 here). In the glassy state, LCT films showed roughly constant modulus (E') values *i.e.* 11-18 GPa,
399 compared to ~ 2 GPa for the neat epoxy film. The observed glass transitions (T_g) of the epoxy resins at
400 about -25 °C were in good agreement with the results of DSC (Table 2). After the glass transition of the
401 neat epoxy film, the storage modulus decreased to a minimum ($E' < 1$ MPa) as the temperature reached
402 ambient conditions. Similarly, the LCT films displayed glass transitions between -30 and -20 °C, but
403 held storage moduli (E') between 300 MPa to 1 GPa at ambient conditions depending on composition.
404 The LCT films also showed slight increases in modulus at higher temperatures, likely caused by
405 additional curing of secondary amines (Ortiz et al., 1998a). Overall, trends in the evolution of storage
406 modulus were similar for all LCT films regardless of the CNCs composition. Nevertheless, it appeared
407 that increased amounts of CNCs in LCT films resulted in increased storage modulus, especially at the
408 high temperature range (> 100 °C). Given that CNCs were the primary component of the LCT films,
409 and well above their percolation threshold (Bras et al., 2011), a highly percolated anisotropic network
410 was the most likely cause for the observed mechanical stability. Overall, it was concluded that LCT
411 films gave desirable thermo-mechanical properties compared to pure CNC films which were too brittle
412 for dynamic mechanical analysis and gave up to a 1000-fold increase in storage modulus (E') compared
413 to the neat epoxy film at ambient conditions.

414 Traditional LCTs, which incorporate ‘frozen’ thermotropic liquid crystals, exhibit high glass
415 transitions temperatures ($T_g > 150$ °C) and high moduli ($E \sim 1$ GPa) at room temperature (Barclay and
416 Ober, 1993; Ortiz et al., 1998a; Shiota and Ober, 1997). On the other hand, liquid crystalline

417 elastomers (LCEs) are more loosely cross-linked networks exhibiting low glass transitions ($T_g < 35$ °C)
418 and low moduli ($E < 1$ MPa) at room temperature (Giamberini et al., 2005; Ortiz et al., 1998b; Ribera
419 et al., 2006). In the latter case, liquid crystalline phase transitions are also observed due to high
420 mobility of mesogens within the network. Since cellulose nanocrystals are lyotropic, the presence of a
421 suitable solvent, in most cases water, is a prerequisite to observe liquid crystalline phase transitions
422 (Habibi et al., 2010; Moon et al., 2011). The LCT films produced in this work were via slow
423 evaporation of water, thus, once dried and cured, CNCs existed in a ‘frozen’ state similar to traditional
424 LCTs. It is important to note that in the case of traditional LCEs and LCTs, mesogens are usually
425 covalently cross-linked within the polymeric network, but in this case, LCT films consisted primarily
426 of non-cross-linked CNCs within a secondary non-mesogenic epoxy network. Interestingly, these
427 cellulose-based LCT films uniquely exhibited thermo-mechanical properties of both classes of cross-
428 linked liquid crystalline networks. Although they displayed very low glass transition temperatures, *i.e.*
429 -25 °C, thus existed in the rubbery state at room temperature similar to LCEs, they also displayed high
430 moduli similar to LCTs, *i.e.* from 300 MPa to 1 GPa (see Figure 5(a)). Of course, the results of DSC
431 (Table 2) showed that pure CNCs had a glass transition (T_g) of 211 °C, but this could not be observed
432 in the LCT films. This demonstrated that cellulose nanocrystals have a unique advantage as the
433 mesogenic components of LCTs, due to their combined high modulus, high T_g and ‘frozen’ LC phases
434 within solid films.

435 The results of DMA measurements were also confirmed by nonlinear deformation (stress vs.
436 strain) experiments conducted on LCT and neat epoxy films at ambient conditions. Typical stress-strain
437 curves are shown in Figure 5(b), in which the insert displays the same example curves from 0 to 8 %
438 strain. The neat epoxy film showed behavior more typical of an elastomer, exhibiting low elastic
439 modulus ($E < 1$ MPa), very low tensile strength and high elongation at break, *i.e.* ~40 %. Contrarily,

440 the LCT films showed significantly higher modulus and lower elongation at break. Typical elastic
441 moduli (*i.e.* Young's modulus) of the LCT films derived from the linear portion of the stress-strain
442 curve were from 700 MPa to 1.5 GPa depending on the CNCs composition, and were in reasonable
443 agreement with results of the **dynamic mechanical** measurements (Figure 5(a)). Also, the elongation at
444 break of LCT films was typically from 2 to 8 %, with ultimate tensile strengths from 5 to 7 MPa. These
445 values were found to be comparable to other types of epoxy LCTs based on thermotropic liquid
446 crystals, even though most traditional epoxy LCTs display much higher glass transitions, thus exist in
447 the glassy state at room temperature (Kannan and Sudhakara, 2011). As compared to epoxy-based
448 LCEs (Giamberini et al., 2005; Ortiz et al., 1998b; Ribera et al., 2006), similar, if not significantly
449 enhanced, modulus values were observed in the cellulose-based LCT films. As mentioned previously,
450 the results of DSC (Table 2) showed that pure CNCs had a glass transition (T_g) of 211 °C, but this
451 could not be observed in the LCT films. In general, it appeared that with high content of CNCs, LCT
452 films exhibited increased elastic modulus, but decreased elongation at break. As previously discussed,
453 CNCs were the primary component of the LCT films, and well above their percolation threshold (Bras
454 et al., 2011), thus a highly percolated anisotropic network was the most likely cause for the observed
455 mechanical stability. The mechanical properties of pure CNC films were very difficult to analyze due
456 to their brittleness. Nevertheless, tensile tests have been successfully performed by Bras et al. on a
457 variety of CNC films from different cellulose sources, in which the experimental elastic modulus (E) of
458 2.13 GPa was determined for a cotton CNC film. This appeared to be in good agreement with the
459 results of our tensile tests. The mechanical properties, such as elastic modulus (E), of LCT films based
460 on CNCs were also increased in comparison to previous studies of epoxy nanocomposites (Ruiz et al.,
461 2000; Ruiz et al., 2001; Rusli and Eichhorn, 2011; Tang and Weder, 2010). This was expected due to
462 the high compositions of CNCs within LCT films (up to 72 % wt.) as compared to previous epoxy

463 nanocomposites which only contained up to 24 % (v/v) CNC reinforcement. In general, LCT films
464 based on renewable cellulose nanocrystals were strong and highly durable, which may provide better
465 use for practical applications.

466

467 *Electron Microscopy*

468 In order to visualize the microstructure of LCT films, images were obtained from thinly sliced
469 sections by transmission electron microscopy (TEM). Sections were taken from various areas of **four**
470 **different films** to confirm the optical heterogeneities observed by polarized light microscopy (Figure
471 2). Shown in Figure 6 are typical examples of textures obtained from LCT films at (a) the edges and (b)
472 near the center. Interestingly, as seen in Figure 6(a), the textures at the film edges appeared to show
473 large microscopic domains of oriented CNCs, as opposed to small and localized domains near the film
474 center, shown in Figure 6(b). **The observed textures at film edges were consistent between four samples**
475 **therefore was not likely an artifact due to sample preparation (microtome)**. Similar LC phases have
476 been observed in the case of solid films of self-assembled CdSe nanorods (Li and Alivisatos, 2003;
477 Talapin et al., 2004). As mentioned previously, the heterogeneities in LCT film microstructure were
478 likely due to a complex synergy between surface energy phenomena associated with water evaporation
479 from hydrophobic surfaces (Picknett and Bexon, 1977), the “coffee ring” effect (Uetani and Yano,
480 2012) and the intermolecular interactions of epoxy monomers with CNCs. The “coffee ring” effect has
481 been utilized to analyze CNC anisotropic structures, in which highly oriented phases occurred at the
482 ring perimeter of evaporated water droplets. Also, since the LCT films were noted to have higher
483 thickness near the center, it was likely that higher concentrations of cured epoxy were present, such
484 that CNCs were able to self-assemble into oriented phases more effectively at the film edges with less
485 disruption from non-mesogenic epoxy monomers. There was no evidence of CNCs assembled into

486 periodic LC phases as observed in pure cellulose nanocrystal films (Picard et al., 2012; Revol et al.,
487 1992), which were likely disrupted by the presence of non-mesogenic epoxy monomers. Nevertheless,
488 the nanostructures shown in Figure 6 confirmed highly localized and microscopic domains of oriented
489 CNCs within the LCT films. The noted structural differences between various locations of LCT films
490 were further evidence that by varying the processing conditions, CNCs liquid crystalline phases could
491 potentially be fine-tuned. This conclusion was supported by the birefringence patterns observed by
492 polarized optical microscopy (Figure 2) and the different reflection intensities of XRD spectra (Figure
493 4) for LCT films.

494

495 **Conclusions**

496 A new class of liquid crystalline thermosets (LCTs) was successfully produced containing
497 lyotropic cellulose nanocrystals as the primary mesogenic component by the addition of non-
498 mesogenic epoxy monomers. These LCTs were produced by totally aqueous processing methods and
499 ultimately cured at elevated temperatures to produce **ordered networks** of ‘frozen’ liquid crystalline
500 phases. X-ray diffraction experiments highlighted the effects of texture of cellulose nanocrystals within
501 LCT films compared to the lyophilized state. All of the processed films exhibited various degrees of
502 birefringence depending on location as observed by polarized optical microscopy. These findings
503 indicated that **opportunities might exist to fine tune the optical properties** of cellulose-based LCT films
504 by varying the processing conditions. In addition, the LCT films selectively attenuated UV
505 wavelengths. These cellulose-based LCTs exhibited highly desirable thermo-mechanical properties and
506 flexibility compared to neat cellulose nanocrystal films, which were too brittle for dynamic mechanical
507 analysis. Interestingly, these cellulose-based LCT films uniquely exhibited thermo-mechanical
508 properties of both classes of cross-linked liquid crystalline networks, namely liquid crystalline

509 thermosets (LCTs) and liquid crystalline elastomers (LCEs). Although they displayed very low glass
510 transition temperatures (~ -25 °C), thus existed in the rubbery state at room temperature similar to
511 LCEs, they also displayed high moduli (~ 1 GPa) similar to LCTs. Pure CNCs showed a glass transition
512 (T_g) of 211 °C, but this was not observed in the LCT films. This demonstrated that cellulose
513 nanocrystals have a unique advantage as the mesogenic components of LCTs, due to their combined
514 high modulus, high T_g and ‘frozen’ LC phases within solid films. The self-assembly of cellulose
515 nanocrystals into microscopic and highly localized domains of oriented phases was confirmed by
516 transmission electron microscopy. The development of liquid crystalline thermosets based on cellulose
517 nanocrystals and totally aqueous processing methods is expected to open more renewable pathways for
518 designing high performance composites with **ordered network structures and unique optical properties**.

519

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525

526 **Supplementary Data**

527 An AFM height image of cellulose nanocrystals produced from this work is available as
528 supplementary data.

529

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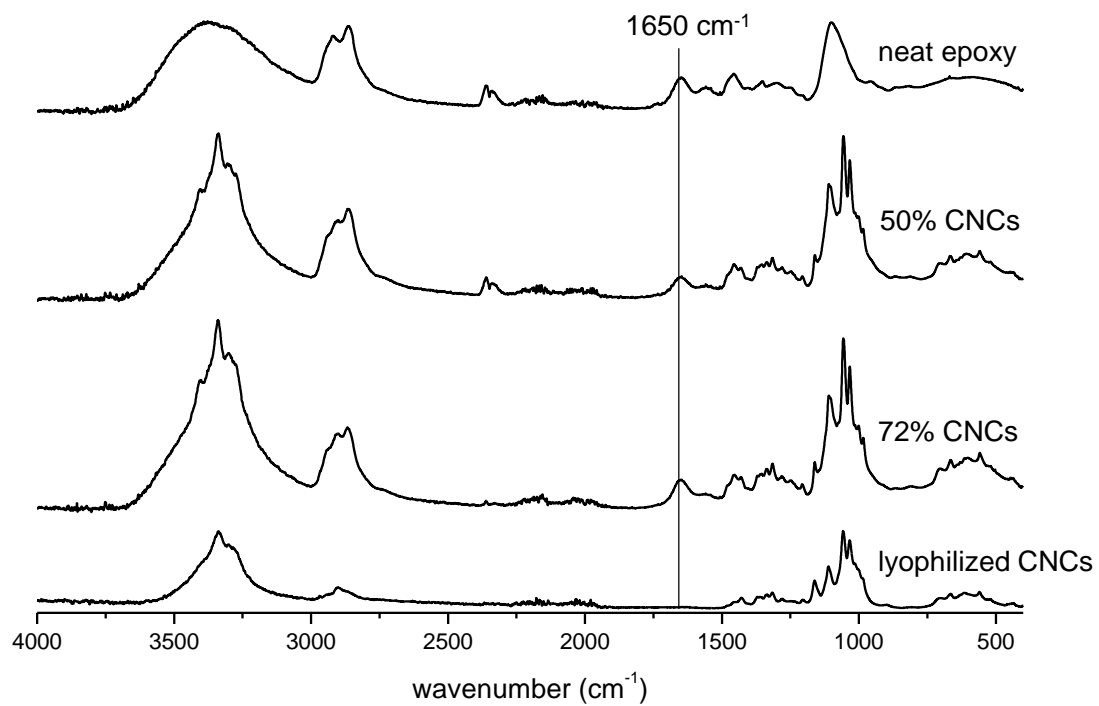


Fig. 1. ATR-FTIR spectra of lyophilized CNCs (bottom), LCT films containing 50 % and 72 % CNCs (middle), and neat epoxy film (top). Note: Line drawn to guide the eye.

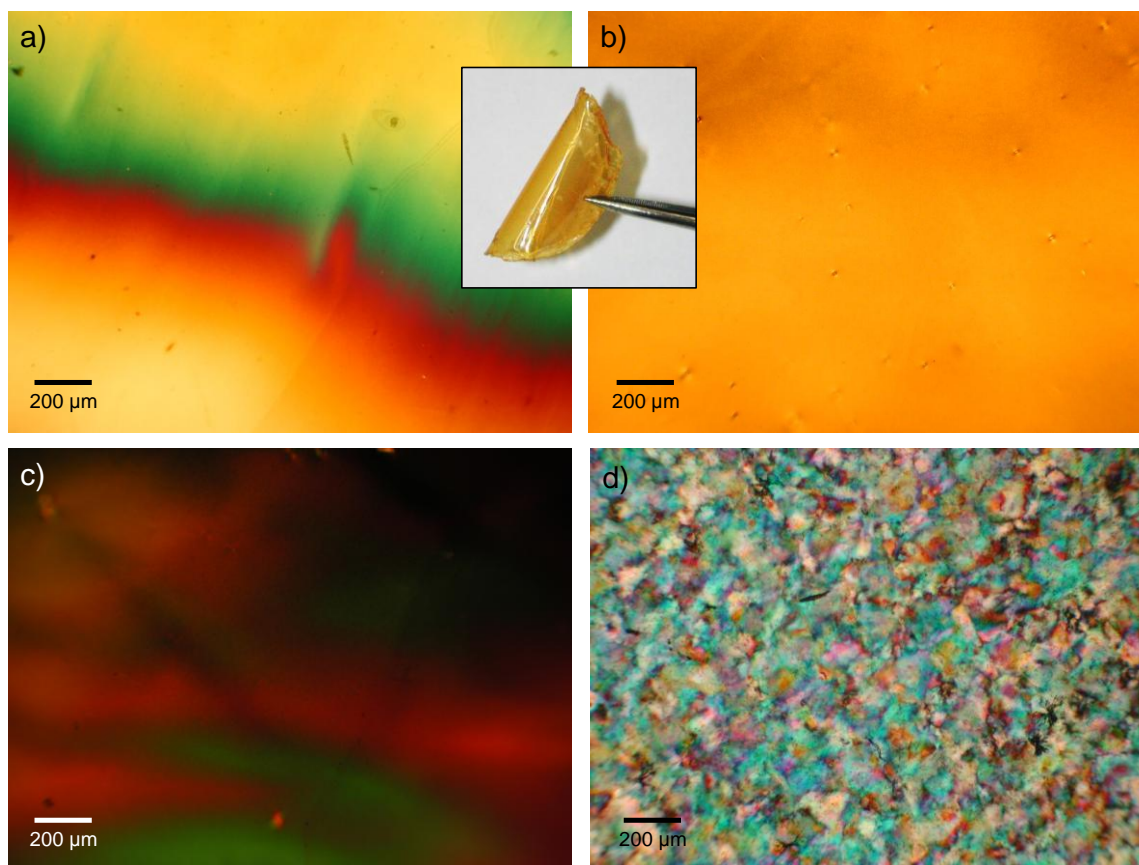


Fig. 2. Typical optical appearance of LCT films (72 % wt. CNCs) at (a) the film edge, (b) ~1 cm from the film edge, (c) the film center under crossed-polarizers, and (d) a pure cellulose nanocrystal film under crossed-polarizers, and (inset) flexed LCT films to the naked eye.

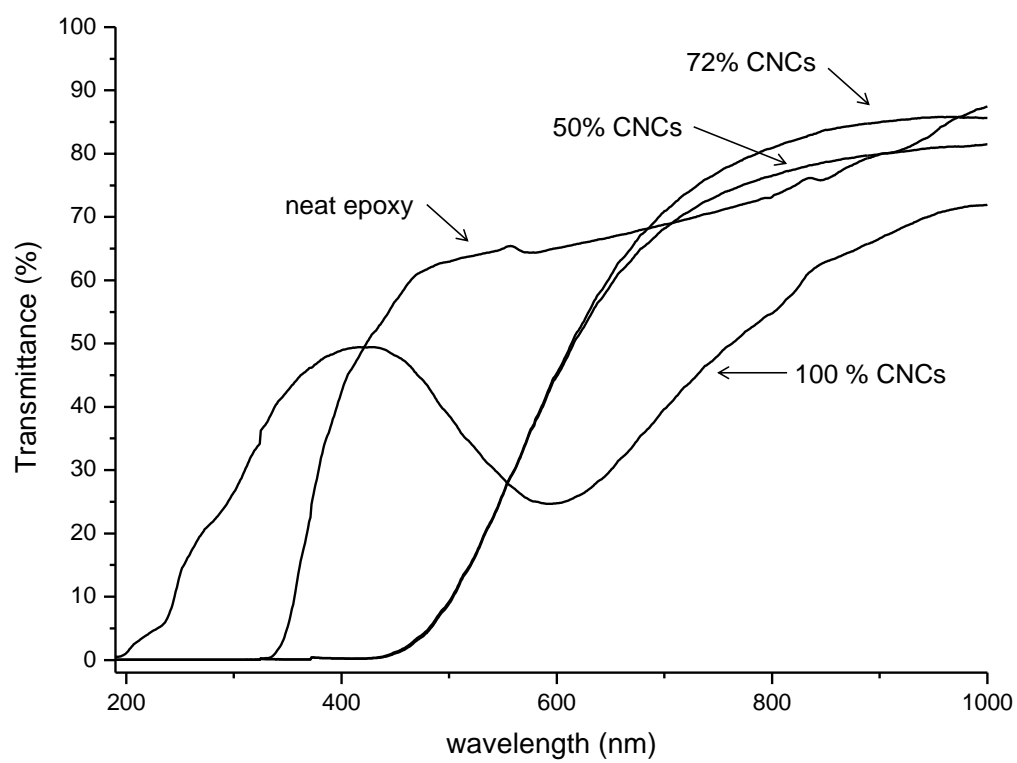


Fig. 3. UV-Vis spectra of a pure CNC film, LCT films containing 50 % and 72 % wt. CNCs, and a neat epoxy film.

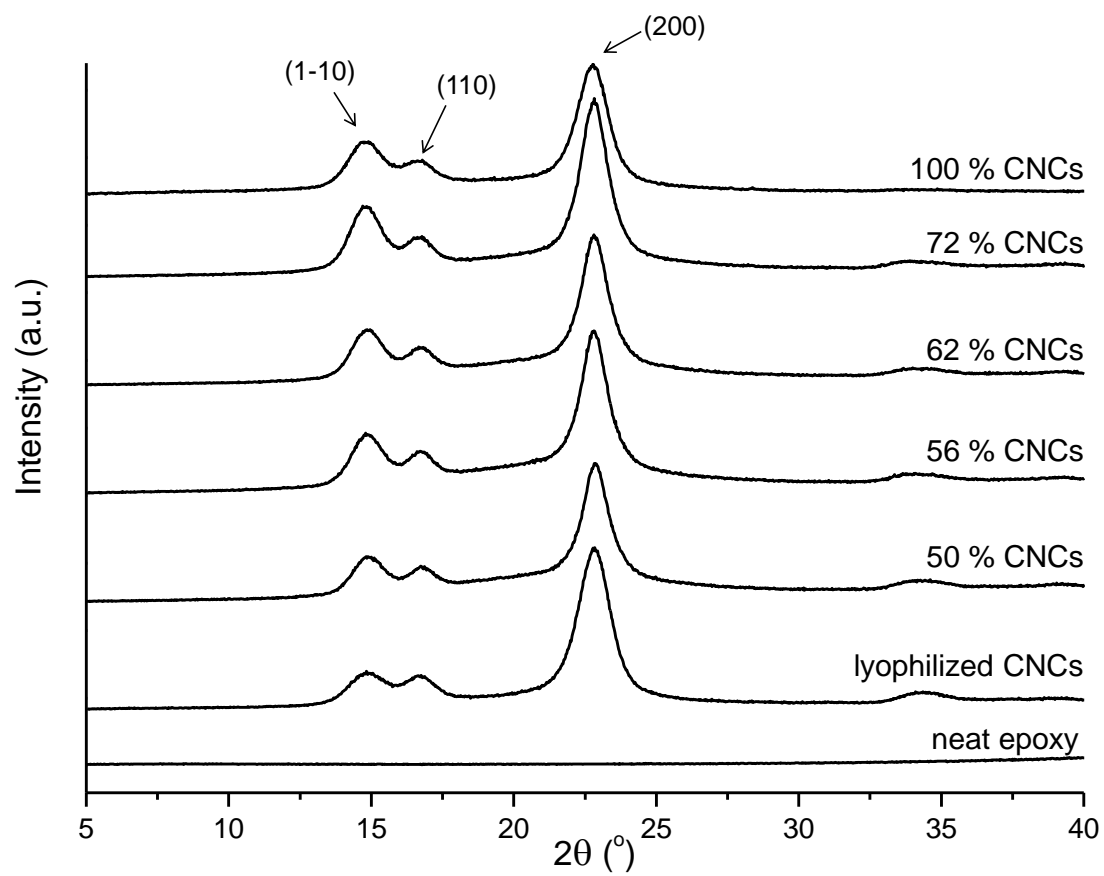


Fig. 4. XRD spectra of a neat epoxy film (bottom), lyophilized CNCs, LCT films containing 50-72 % wt. CNCs and a pure CNC film (top).

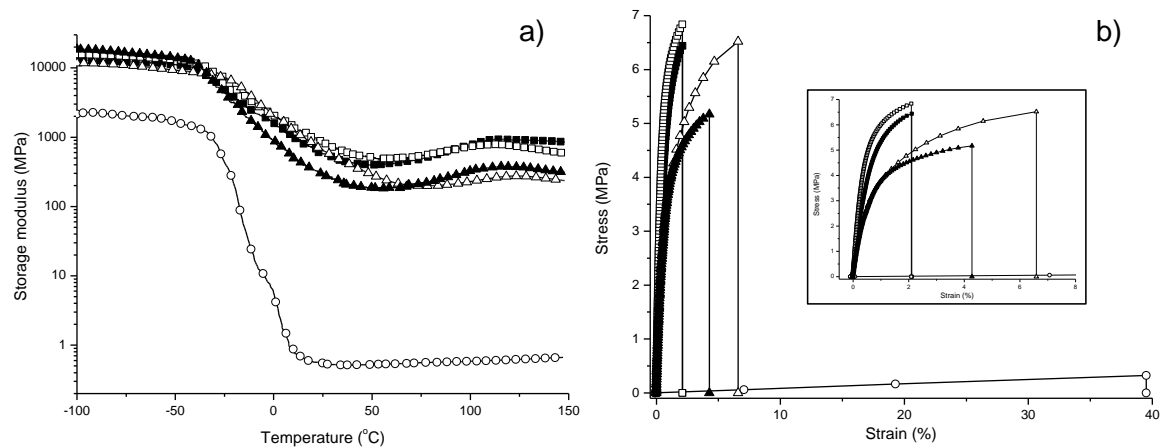


Fig. 5. (a) Evolution of storage modulus (MPa) versus temperature (°C) and (b) typical stress versus strain curves for LCT films containing 50 % wt. CNCs (-▲-), 56 % wt. CNCs (-Δ-), 62 % wt. CNCs (-□-), 72 % wt. CNCs (-■-) and neat epoxy films (-○-). Notes: Logarithmic scale of the y-axis in (a). Inset plot in (b) contains the same curves from 0 to 8 % strain.

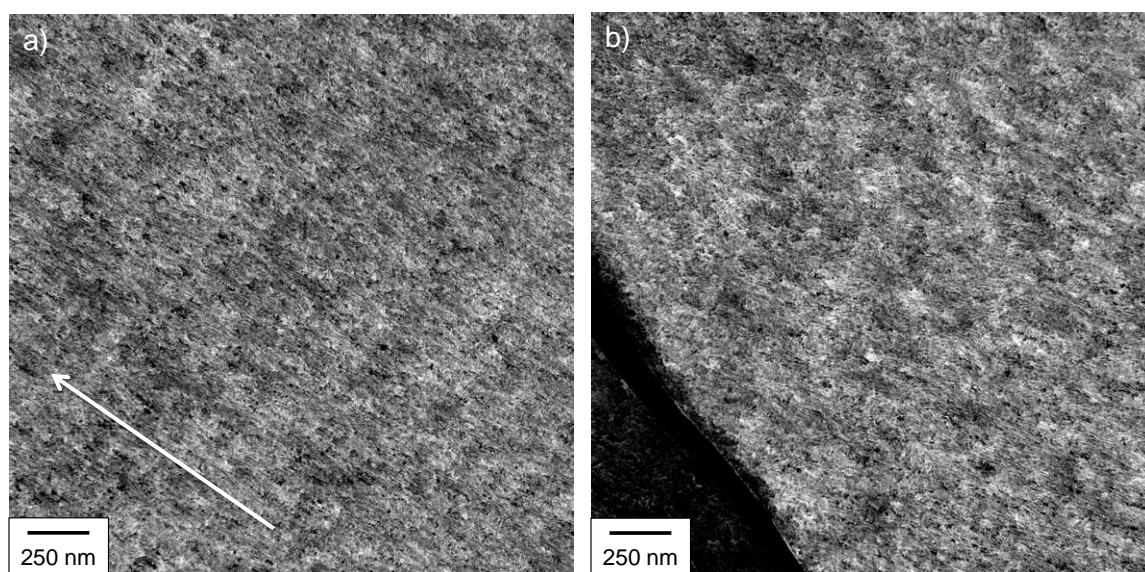


Fig. 6. Typical transmission electron micrographs of LCT films (72 % wt. CNCs) at the (a) edge and (b) center of the film. Notes: Arrow drawn to guide the eye. Dark area in (b) is from the TEM grid.

Table 1. Average crystallite width (D_{hkl}), crystallinity index (CrI) and intensity ratio (I_{200}/I_{1-10}) of samples derived from XRD spectra.

<i>Sample</i>	<i>Crystallite width (nm)^a</i>	<i>Crystallinity index (%)</i>	<i>Intensity ratio (I_{200}/I_{1-10})</i>
Lyophilized CNCs	6.6	90.7	4.0
100% CNCs film	6.4	85.6	2.3
72% epoxy CNCs	7.0	89.3	2.5
62% epoxy CNCs	6.6	86.0	2.6
56% epoxy CNCs	6.5	84.5	2.7
50% epoxy CNCs	6.2	85.1	3.0
neat epoxy	-	-	-

^acalculated by Scherrer's formula.

Table 2. Summary of thermal transitions of CNCs, LCT films and neat epoxy film from DSC thermograms.

<i>Sample</i>	<i>Glass transition (T_g) ($^{\circ}\text{C}$)</i>	ΔC_p ($\text{J g}^{-1} \text{ }^{\circ}\text{C}^{-1}$)	<i>% CNCs</i>	<i>% epoxy</i>
Lyophilized CNCs	211	0.395	100	-
72% CNCs	-	-	-	-
62% CNCs	-23.6	0.320	55	45
56% CNCs	-22.3	0.391	45	55
50% CNCs	-22.8	0.404	43	57
neat epoxy	-21.9	0.706	-	100