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1 **Assessment of chemical and mechanical**
2 **behavior of bamboo pulp and nanofibrillated**
3 **cellulose exposed to alkaline environments**

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32 Abstract

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34 This study was performed to study the effects of the cement paste composition (calcium aluminate
35 cement -CAC- and a geopolymer in comparison to Portland cement -OPC-) on bamboo pulp and
36 nanofibrillated cellulose (NFC). The changes in the composition and chemical structure of the
37 fibers were analyzed by X-ray photoelectron spectroscopy (XPS) and Fourier-transform infrared
38 spectroscopy (FTIR). The changes in the mechanical strength were evaluated through tensile tests
39 on the fibers after immersion on the cement pastes, in the form of sheets. The XPS results showed
40 that the immersion of the pulp and NFC in the different pastes (CAC, geopolymer and OPC)
41 modified the chemical surface of these fibers: it was found removal of lignin and extractives and
42 some degradation of hemicellulose and cellulose. The FTIR analysis indicated modifications in the
43 hydrogen bonds energy. The tensile strength of pulp sheets decreased in 70% and 34% after
44 immersion in OPC and geopolymer, respectively. The tensile strength of the NFC sheets decreased
45 36%, 68% and 54% after immersion in OPC, CAC and geopolymer, respectively. Thus, the
46 response of the bamboo pulp and NFC immersed in different cement pastes was different due the
47 inherent characteristics of such fibers, and not only the Portland cement should be considered as
48 harmful to lignocellulosic fibers. Although CAC and geopolymer are free of calcium hydroxide,
49 the high alkalinity of these pastes also accelerated the degradation process of lignocellulosic fibers.

50 *Keywords:* bamboo pulp, nanofibrillated cellulose, degradation, high alkalinity

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66 1. Introduction

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68 Fiber-cement composites are construction building materials that can be used in
69 applications such as sidings, roofing and ceiling panels. These materials are basically composed of
70 Ordinary Portland cement (OPC), limestone, cellulosic pulp and synthetic fibers, like
71 polypropylene, and polyvinyl alcohol. In the Hatschek process cellulosic pulps are used as
72 processing fiber. In this process the pulps are responsible for the retention of particles in the
73 cementitious matrix and to facilitate the water removal during the drainage stage of the production
74 process. In addition, the pulp is also a reinforcing component in the micrometric scale of fiber-
75 cement (Bentur and Mindess 2007).

76 Other form of cellulosic fibers with potential to use as reinforcement of cement-based
77 materials is nanofibrillated cellulose (NFC). NFC is produced by mechanical disintegration
78 without chemicals or even biological process. It is described as a long and flexible cellulosic nano-
79 material, which is obtained from cellulose fibers that have been fibrillated to achieve agglomerates
80 of cellulose microfibril units (Missoum et al. 2013). NFCs contain crystalline and amorphous
81 regions and they have diameters in the range of 5–60 nm and lengths of several hundred
82 nanometers (Klemm et al. 2011; Klemm et al. 2018). NFCs have high specific surface area, which
83 increase the interaction of the fibers with the matrix promoting the higher mechanical strength of
84 the reinforced composites (Correia et al. 2018a; Correia et al. 2016; Ardanuy et al. 2012a;
85 Ardanuy et al. 2012b). According to Correia et al. (2018b) NFC contribute to the increase of the
86 physical and chemical adhesion, friction and mechanical anchorage with the matrix with respect
87 pulps, induced by its high specific surface area.

88 Eucalyptus and pinus pulp are traditionally used in the producing of commercial fiber-
89 cement products by the Hatschek process, but some studies have shown that bamboo pulp and
90 NFCs also have potential to be used as reinforcement for these cement-based materials (Correia et
91 al. 2018a ; Li et al. 2017; Xie et al. 2015; Correia et al. 2014; Rodrigues et al. 2010; Rodrigues et
92 al. 2006; Coutts and Ni 1995).

93 The drawback of the use of lignocellulosic fibers in cementitious matrices based on OPC
94 is their low durability in this environment. The high alkalinity ($\text{pH} \approx 13$) and the presence of

95 calcium hydroxide in the pore solution of OPC affect the durability of cellulosic fibers (Cheng et
96 al. 2018; Wei 2018; Wei and Meyer 2015; Tolêdo Filho et al. 2003).

97 The high alkalinity of water in the pore of the cementitious matrix dissolves
98 the lignin and hemicelluloses of fiber; alkaline hydrolysis of cellulose molecules leads to
99 degradation of molecular chains and the reduction in the degree of polymerization and
100 lower tensile strength; crystallization of calcium hydroxide in the lumen, walls of the individual
101 fibers and middle lamella leads to a decrease in the fiber flexibility and mechanical strength (Singh
102 1985; Gram 1988; Tolêdo Filho et al. 2000; Mohr et al. 2005; Wei and Meyer 2017). Thereby, the
103 degraded lignocelulosic fibers can decrease the durability of the material and consequently lead it
104 to premature failure (Wei and Meyer 2017).

105 The use of cellulosic pulp as reinforcement of cement-based materials is a way to
106 mitigate the degradation because the high alkalinity of cementitious matrix deteriorates the lignin
107 and hemicellulose, and in chemical pulping processes most part of these components is removed.
108 Other measures to reduce fiber degradation and increase the durability of composites include the
109 use of calcium hydroxide free cementitious matrices or the use of supplementary cementitious
110 materials as a partial replacement of OPC (Tolêdo Filho et al. 2003; Mohr et al. 2007) and the use
111 of accelerated carbonation curing (Almeida et al. 2013; Santos et al. 2015). According to Mármol
112 and Savastano (2017), these measures aim to reduce the level of calcium hydroxide in cement
113 matrices.

114 Gram (1988) studied the durability of sisal fibers in matrix with high alumina cement and
115 concluded that the use of these cements significantly minimize the degradation of composite
116 subjected to accelerated aging. In this study the author observed that after the alternating of 120
117 cycles of immersion and drying, the post-cracking peak strength of composite with high alumina
118 cement decreased by 56.2% in comparison to 98.8% for the OPC matrix. According to Mohr et al.
119 (2007) this is due the consumption of calcium and aluminate ions from the pore solution that
120 decreases the amount of calcium hydroxide and secondary ettringite reprecipitation.

121 Claramunt et al. (2018) evaluated the mechanical behavior of calcium aluminate cement
122 (CAC) composites reinforced with flax's nonwoven fabrics, and the durability of the composites
123 subjected to five cycles of immersion and drying. The authors found that the CAC matrix is
124 effective for mitigating the degradation of the fibers after accelerated aging. These results are in
125 accordance with Mohr et al. (2005), who reported that CAC is an alternative to supplementary

126 cementitious materials and should be evaluated for their respective effect on the progression of
127 degradation of pulp in the fiber-cement composite.

128 Another promising matrix rich in aluminates and free of calcium hydroxide is the
129 geopolymer. Among the advantages of the use of geopolymers as an inorganic matrix is that this
130 matrix is considered eco-friendly. The cure of geopolymers is generally performed at room
131 temperature and, therefore, its synthesis requires less energy, and depending on the activator there
132 is a reduction of up to 80% CO₂ compared to OPC (Tailby and MacKenzie 2010; Villa et al.
133 2010; Firdous et al. 2018).

134 Due to fragile behavior of geopolymer, the use of vegetal fibers as reinforcement of this
135 matrix has been studied (Alomayri et al. 2013; 2014; Chen et al. 2014; Assaedi et al. 2015).
136 However, as the high alkalinity of the geopolymer can be harmful to the lignocellulosic fibers, Yan
137 et al. (2016) suggest the composition of the geopolymer with high content of fly ash, blast-furnace
138 slag, metakaolin and silica. But there are no studies to confirm the increased durability of the
139 fibers and geopolymer composites with the adoption of these measures.

140 Thus, this study aims assess the chemical and mechanical degradation of bamboo pulp
141 and nanofibrillated cellulose when exposed to three different highly alkaline unhardened cement
142 pastes (Portland cement, calcium aluminate cement and geopolymer) up to 28 days of age, what
143 can be considered a very severe degradation test.

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145 **2. Experimental procedure**

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147 **2.1 Preparation and characterization of the fibers**

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149 Unbleached pulp and nanofibrillated cellulose (NFC) from bamboo were used in this
150 study. The pulp was produced by the organosolv pulping process using a batch reactor. The NFC
151 was produced from bamboo pulp using the grinding method. The parameters used for the pulping
152 and nanofibrillation, as well as their characterization are described in Correia et al. (2016). The
153 chemical, physical and morphological characteristics of the fibers are presented in Table 1.

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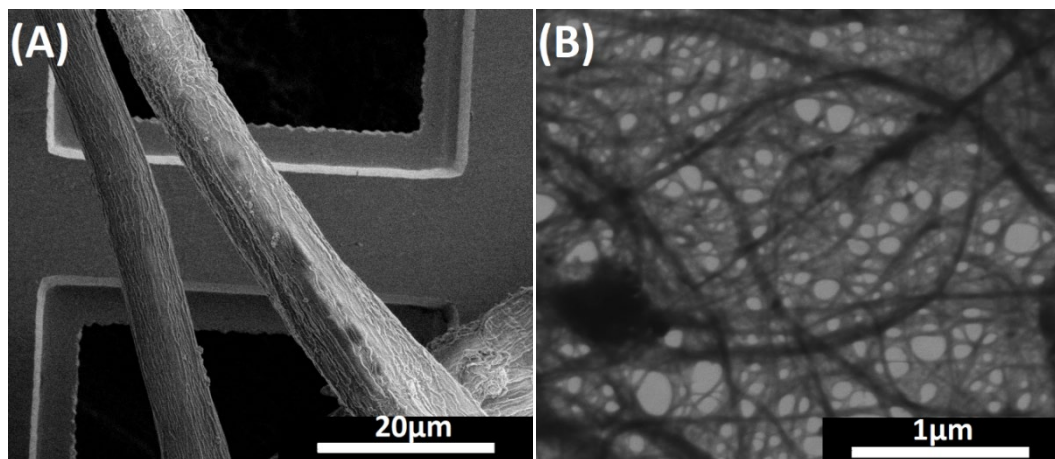
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Table 1 – Chemical, morphological and physical characteristics of the pulp and NFC

Chemical composition		
Components (%)	Pulp	NFC
Extractives	6.92 ± 1.47	9.07 ± 1.25
Lignin	9.85 ± 3.86	9.67 ± 2.37
Holocellulose	85.39 ± 1.42	85.28 ± 1.96
Cellulose	82.75 ± 0.42	77.82 ± 1.02
Hemicellulose	2.64 ± 0.42	7.46 ± 1.02
Morphological and physical characteristics		
Average diameter	(2.72 ± 2.64) μm	(13.88 ± 9.47) nm
Density (g/cm ³)	1.51	1.42

156

157 The micrographs obtained via Scanning Transmission Electron Microscopy (STEM) (FEI
 158 Magellan 400 L Scanning Electron Microscope) are presented in Figure 1. The images show the
 159 difference between pulp and NFC related to their morphology and specific surface area. In this
 160 way, considering the difference in the levels of some chemical components (extractives, cellulose
 161 and hemicelluloses) and in the morphology is possible to presume that the response of these fibers
 162 immersed in different alkaline environment will not be the same.



163

164 Figure 1 – Micrographs of unbleached bamboo pulp (A) and NFC (B) used in the degradation
 165 study

166

167 **2.2 Composition of the alkaline pastes**

168

169 In order to analyze the degradation of the fibers in different alkaline environments,
170 samples of pulp and NFC were prepared in packets, which were produced using a first layer of
171 paper filter and other layer of fabric material. Then, the packets with fibers were immersed in three
172 containers containing the unhardened pastes: Ordinary Portland Cement (OPC) CEM I 52.5 R (EN
173 197-1: 2011 Standard); Calcium Aluminate Cement (CAC) (EN 14647 Standard), both with a
174 water:cement ratio of 2:1; and geopolymer. The OPC and CAC were supplied by Molins
175 industrial, *Molins de Rei*, Barcelona, Spain. The geopolymer was produced using metakaolin as
176 aluminosilicate source. The activator solution for geopolymerization was made by mixing sodium
177 hydroxide (10M) with silica fume. The metakaolin and silica fume were supplied by Arciresa,
178 Gijón, Asturias, Spain. The molar ratios used were Si/Al = 1.72, Na/Al = 1.01, and H₂O/Na₂O =
179 11.25.

180 The objective of this methodology was keeping the fibers in a high alkaline environment,
181 and in contact with the ions of the unhardened cementitious pastes, without direct contact between
182 fibers and mineral particles. After immersion of the packets with fibers in the cement pastes, it was
183 intermittently stirred during 7 days at 1000 rpm to avoid cements hardening. In order to monitor
184 the evolution of the alkalinity of unhardened pastes up to 28 days, which is very severe condition
185 that the fibers were exposed, the pH of the pastes was measured using a bench top pH meter at
186 different ages (0 h, 6 h, 7 days and 28 days). The packets were kept within unhardened pastes for
187 28 days.

188 The chemical composition (%wt of oxides) of the OPC, CAC and mineral particles of
189 geopolymer (metakaolin and silica fume) was analyzed by X-ray Fluorescence Spectrometry, and
190 the results are presented in Table 2.

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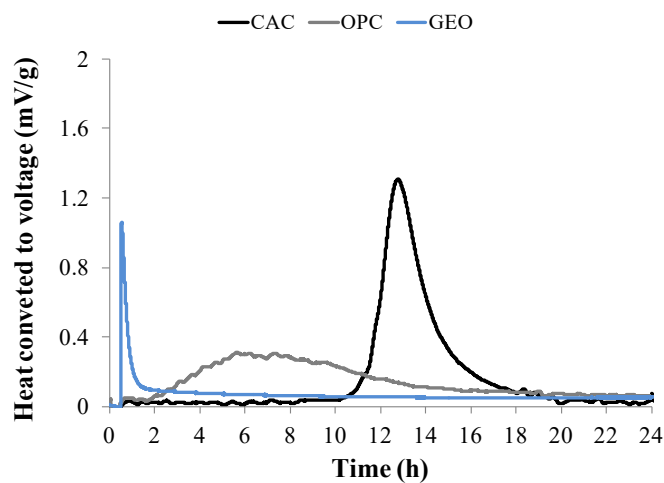
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197 Table 2. Chemical composition (%wt of oxides) of Portland cement (OPC), calcium aluminate
 198 cement (CAC), metakaolin and silica fume

Oxides (%)	OPC	CAC	Metakaolin	Silica fume
SiO ₂	19.26	3.7	54.61	93.5
Al ₂ O ₃	4.02	41.5	40.24	0.32
CaO	57.32	38.1	0.1	0.39
MgO	5.04	0.43	-	0.2
SO ₃	3.62	0.01	-	0.03
Na ₂ O	0.14	0.14	-	0.23
K ₂ O	1.15	0.02	0.51	0.28
TiO ₂	-	1.72	0.13	-
MnO	-	0.01	-	-
Fe ₂ O ₃	3.91	13.20	0.52	0.11
Loss on ignition	2.98	1.41	0.07	3.44

199

200 The heat of hydration of the matrices based in OPC, CAC and geopolymer were
 201 determined using the calorimetry test. The pastes with a water/cement relation of 0.5 were mixed
 202 during 5 min at temperature of 25 °C, and then they were placed in glass cylinders. The sealed
 203 samples were placed in the calorimeter, which recorded the heat of hydration and reaction during
 204 24 h. Figure 2 shows the heat of hydration of OPC and CAC and the heat of reaction of
 205 geopolymer.



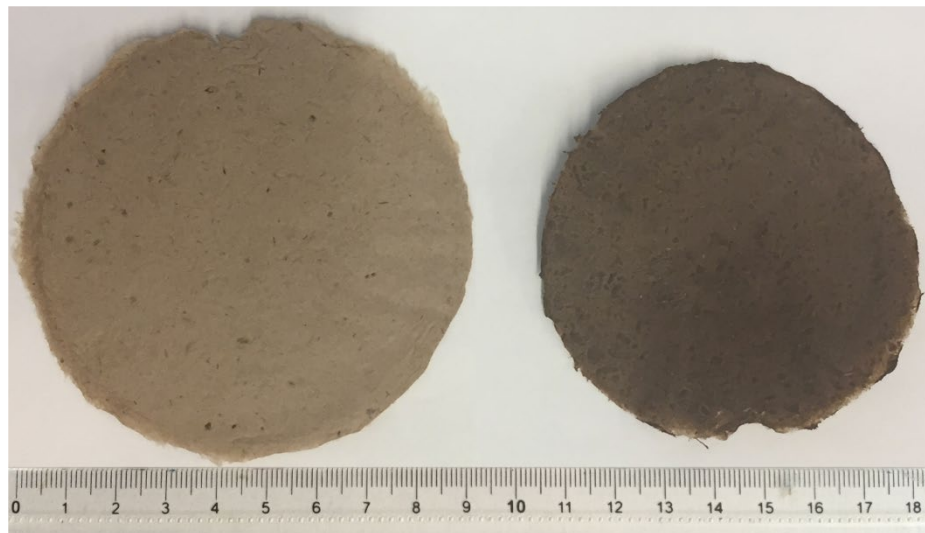
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207 Figure 2 - Curves of the heat of hydration of OPC, CAC and reaction of geopolymer

208 2.3 Study of fibers degradation

209

210 Sheets of pulp and NFC immersed in the unhardened pastes with OPC, CAC and
211 geopolymer at 28 days were produced for further characterization and analysis of fibers
212 degradation. The sheets of 0.05 mm of thickness were produced by the vacuum filtration from the
213 fibers suspension with 0.2% consistency (by mass). The suspensions were stirred in a laboratory
214 mixer during 5 min and then, they were filtered using hydrophilic polyether sulfone membranes
215 with 90 mm of diameter and pores of 0.1 μm of thickness. After filtration, the sheets were
216 removed from the membrane and then they were oven dried at 60 °C for 48 h. Figure 3 presents
217 the sheets of pulp and NFC that were immersed in alkaline environments.



218

219 Figure 3– Sheets of pulp (left) and NFC (right) produced after immersion in the alkaline
220 environment

221

222 The changes in the chemical structure of the fibers before and after the accelerated
223 degradation test, from immersion of the pulp and NFC in the three alkaline pastes, were analyzed
224 by the X-ray photoelectron spectroscopy (XPS) and Fourier-transform infrared (FTIR)
225 spectroscopy.

226 The XPS measurements were performed using a SPECS system equipped with Al anode
227 XR50 source operating at 150 W and 12 kV and a Phoibos 150 MCD-9 detector. Samples were
228 placed in a vacuum chamber (5.0×10^{-8} mbar). Survey spectra were recorded in 0.1s/step, 0.5eV
229 /step and 5 scans. The high-resolution were recorded in 0.5 s/step, 0.05eV/step and 3 scans.

230 Spectra were analyzed using SPECS Prodigy and data processing were performed with CasaXPS
231 software.

232 The FTIR spectra of pulp and NFC were recorded using a Thermo Scientific Nicolet 6700
233 FTIR instrument equipped with an Attenuated Total Reflectance (ATR) Smart Orbit Diamond
234 accessory that allows collection of FTIR spectra directly on a sample. The spectra were obtained
235 over the scan range of 4000–600 cm^{-1} , with a spectral resolution of 4 cm^{-1} , and with a total
236 accumulation of 32 scans. Curve fitting for the peak deconvolution was performed by OriginPro
237 2018 software. The true shape of the peaks obtained from the absorption bands for samples was
238 assumed to be Lorentzian. The number of peaks involved was determined in the range of 2700-
239 3800 cm^{-1} .

240 The energy of the hydrogen bonds was calculated using the Equation 1 (Poletto et al.
241 2014; Popescu et al. 2007):

$$242 \quad E_H = \frac{1}{k} \left[\frac{v_0 - v}{v_0} \right] \quad (1)$$

243 Where v_0 is the standard frequency corresponding to free OH groups (3650 cm^{-1}), v is the
244 frequency of the bonded OH groups, and k is a constant ($1/k = 2.625 \times 10^2 \text{ kJ}$).

245 The sheets produced for the mechanical characterization were cut in specimens with
246 approximately 30 mm of length and 5 mm of width. For each type of cementitious paste, 22
247 specimens of NFC and 28 specimens of pulp were produced for the mechanical test. The
248 specimens were subjected to tensile testing using a texture analyser model TA XT Plus with a load
249 cell of 5 N, the cross head speed of 0.1 mm/s and with the gage length of 10 mm.

250 A statistical analysis was carried out for comparison the tensile strength of fibers
251 immersed in OPC, CAC and geopolymer with the control. The statistical analysis was performed
252 using Statistica software (version 13.4.0.14). After the analysis of ANOVA assumption, an
253 assessment was defined by completely randomized design (CRD) and a comparison between the
254 average values by Tukey test at 5% significance level.

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258 3. Results and discussions

259

260 The pH values of the unhardened cementitious pastes at different ages are presented in
261 Table 3. The slight increase in pH of the OPC and CAC pastes over time was observed. This is due
262 the dissolution of minerals and continuous cement hydration reactions, which forms new hydration
263 products with water (Šiler et al. 2016).

264 Sodium hydroxide is commonly used as alkaline activator of the silicon and aluminum for
265 the geopolymer synthesis. The use of hydroxides and alkali silicates with pH values higher than 13
266 allows transforming glassy structure partially or totally into a very compacted composite (Khale
267 and Chaudhary 2007). The apparent reduction of the pH of the geopolymer, especially in 6 h, may
268 indicate the dissolution of the silicate and the aluminate and the polymerization of the material,
269 since the gain of 70% strength of the geopolymer occurs in the 3-4 h of cure (Khale and
270 Chaudhary 2007).

271 The results show that despite the changes occurred in the pH of the pastes over the 28
272 days, such matrices remained highly alkaline.

273

274 Table 3 – pH evolution over time of OPC, CAC and geopolymer

Cement Age	OPC	CAC	Geopolymer
0 h	12.95	11.81	13.37
6 h	12.96	11.86	12.84
7 days	13.05	12.46	12.75
28 days	13.16	12.45	12.44

275

276 The analysis of the fibers by XPS allowed the evaluation of the effect of the fibers
277 immersion in different unhardened cement pastes and the changes in the composition and chemical
278 structure of the fibers surface. The chemical composition found on the surface of the fibers is
279 shown in Table 4.

280 Table 4 shows that more than 90% of the fibers surfaces are composed for carbon and
281 oxygen. Besides carbon and oxygen, low amount of nitrogen and calcium was found on the surface

282 of the fibers. The higher content of calcium is observed, especially, in the fibers (pulp and NFC)
 283 immersed in the OPC paste and in the pulp immersed in the CAC paste, in comparison to the
 284 control fibers. This is due to the residual calcium ions existing in these cementitious pastes, which
 285 migrated to the surface of the fibers. The study of the chemical components on the surface of the
 286 fibers is commonly carried out from the oxygen to carbon (O/C) ratio and determination of the
 287 non-oxygenated carbons in the high resolution spectrum C1s (Johansson et al. 2005).

288

289

Table 4 – Chemical composition and O/C ratio of pulp and NFC surfaces

Fiber		Chemical elements				
		% C	% O	% N	% Ca	O/C
Control	Pulp	72.12 ± 1.30	26.89 ± 1.19	0.70 ± 0.11	0.29 ± 0.03	0.37 ± 0.02
	NFC	71.81 ± 0.33	27.12 ± 0.82	0.66 ± 0.04	0.41 ± 0.05	0.38 ± 0.01
OPC	Pulp	70.33 ± 1.27	28.21 ± 1.25	0.96 ± 0.15	0.51 ± 0.05	0.40 ± 0.01
	NFC	72.96 ± 0.25	25.71 ± 0.70	0.64 ± 0.02	0.69 ± 0.08	0.35 ± 0.01
CAC	Pulp	69.59 ± 1.45	29.21 ± 1.30	0.79 ± 0.12	0.41 ± 0.04	0.42 ± 0.02
	NFC	70.01 ± 0.32	29.08 ± 0.89	0.55 ± 0.03	0.36 ± 0.04	0.41 ± 0.01
Geopolymer	Pulp	69.96 ± 1.36	28.93 ± 1.34	0.84 ± 0.09	0.26 ± 0.02	0.41 ± 0.02
	NFC	70.96 ± 0.42	28.06 ± 0.86	0.58 ± 0.01	0.39 ± 0.05	0.39 ± 0.01

290

291 The O/C ratio is commonly used for the estimation of fibers surface coverage by lignin,
 292 and this is possible based on the theoretical O/C ratios of cellulose and lignin, which are 0.83 and
 293 0.33 respectively (Laine et al. 1994). The results of the chemical composition (Table 1) show that
 294 after nanofibrillation the fibers are composed of about 9% residual lignin, which is confirmed by
 295 the O/C ratio of pulp and NFC control (Table 4). The O/C ratio of all fibers in different conditions
 296 is inferior to 0.83 and close to the O/C ratio of lignin.

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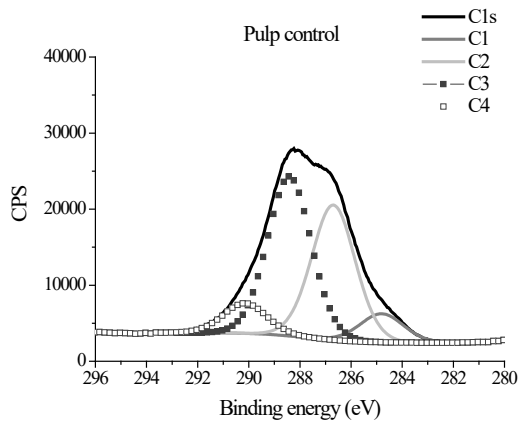
Moreover, the results show that the immersion of the pulp and NFC in the different
 environments modified the chemical surface of these fibers. According to Hua et al. (1993), the
 higher O/C ratio represents the lower amount of lignin and extractives on the surface of the fiber.
 Thus, the immersion of the pulp in the pastes containing OPC, CAC and geopolymer caused the
 removal of lignin and extractives from the surface of the fibers.

302 Regarding to NFC, the increase in the O/C ratio of the fibers immersed in CAC paste and
303 geopolymer was also observed, indicating the removal of lignin and extractives. However, the O/C
304 ratio of NFC immersed in OPC was reduced, indicating a higher amount of lignin on the surface of
305 this fiber. According to George et al. (2015) the degradation of hemicelluloses and removal of
306 extractives of the fibers results in lignin exposure, and consequently the reduction in O/C ratio.

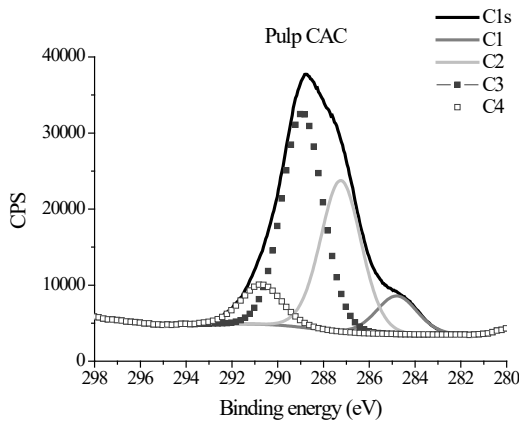
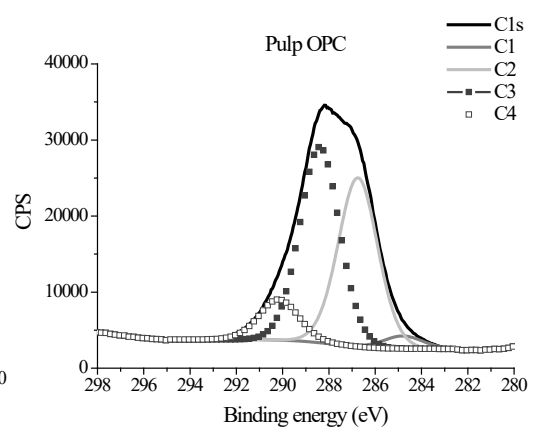
307 The modification of the chemical composition of the fibers immersed in OPC, CAC and
308 geopolymer confirms what Wei and Meyer (2017) and Chakar and Ragauskas (2004) reported.
309 According to the authors the high alkalinity environment causes depolymerization and
310 fragmentation of lignin. This fragmentation leads to the generation of free phenolic hydroxyl
311 groups, and in turn the lignin fragments are rendered water/alkali soluble. Similar with
312 lignin, hemicelluloses also experience alkaline attacks and can be converted into fermentable
313 sugar. Thus, lignin acts as a barrier, being the first pulp component to suffer alkaline attack (Mohr
314 et al. 2005 ; Mohr et al. 2006).

315 High-resolution XPS spectra, which give details of the peak C1s, allow the quantitative
316 and qualitative evaluation of the functional groups in the surface of the fibers from decomposition
317 of the C1s peak in carbon atoms (C1, C2, C3, C4). According to Fuentes et al. (2013) and
318 Migneault et al. (2015), the carbon C1 corresponds to the functional groups C-C and C-H; the C2
319 corresponds to C-O-C and C-OH; the C3 corresponds to C=O and O-C-O; the C4 corresponds to
320 O-C=O. The C1 and C4 components arise mainly from lignin and extractives and C2 and C3
321 components arise mainly from hemicelluloses and cellulose (Hua et al. 1993).

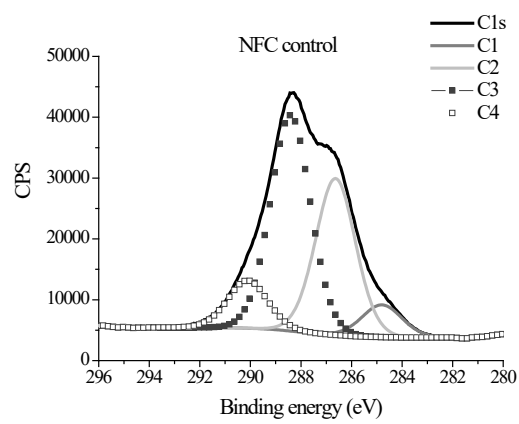
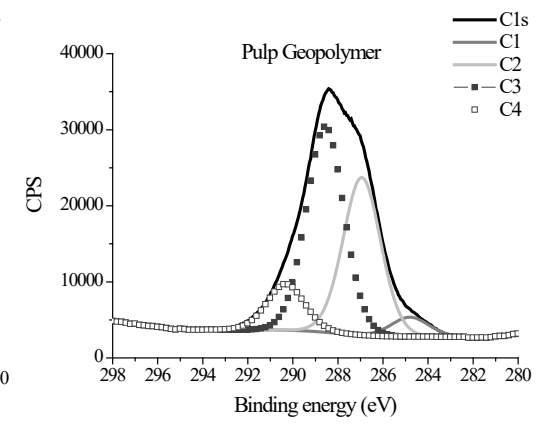
322 The results of the deconvolution of peaks C1s of the pulp and NFC immersed in the
323 different highly alkaline and unhardened cement pastes are presented in Figure 4. Figure 4
324 indicates the existence of the four carbon atoms, C1, C2, C3, C4, for the fibers under all
325 conditions. However, Table 5 shows that the proportion of the atoms and functional groups differs
326 for the pulp and NFC immersed in the different pastes.



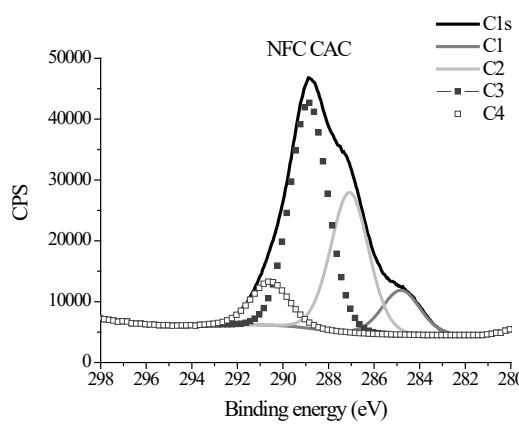
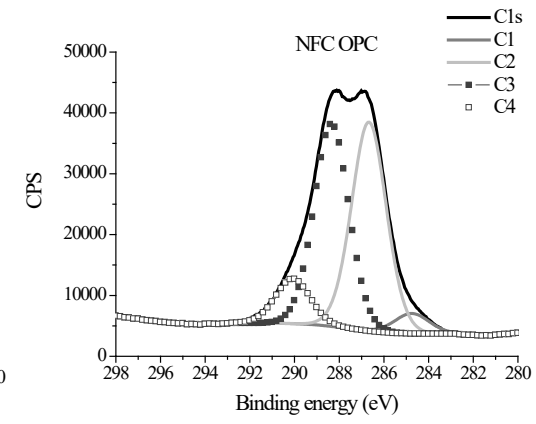
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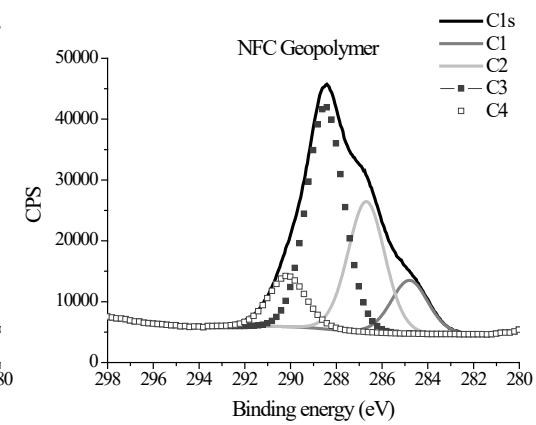
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331 Figure 4 – High resolution XPS scan of C1s region for pulp and NFC immersed in unhardened
332 cementitious pastes based in OPC, CAC and geopolymer

333

334 Table 5 shows that the main modifications occurred in the components C1 and C2 of the
335 pulp and NFC after the immersion in the different alkaline environments, in comparison to the
336 control. In addition, these components are the main ones for the analysis of lignin and cellulose on
337 the surface of the fibers. According to Johansson (2002) the C-C component in C1 spectrum,
338 originating from carbon atoms that have no oxygen neighbors is due to lignin only, but with the
339 presence of extractives, indicated by the C-H bonds. Thus, Johansson (2002) recommends that the
340 quantification of lignin be based on the C-C percentages. For pure cellulose a large contribution of
341 C2 (83%), and a small contribution of C3 (17%) carbons in the C(1s) peak is expected (Hua et al.,
342 1993; Holmberg et al., 1997; Bastidas et al. 2005). This indicates that the C2 component
343 represents much more cellulose, although the hemicellulose has a similar carbon bond structure to
344 cellulose, that is, all the carbons in hemicellulose are linked to at least one oxygen (Hua et al.
345 1993). However, hemicellulose has carbonyl groups (C = O), which are strongly present in the C3
346 and C4 components. In addition, the carbonyl groups are in hemicellulose, lignin and extractives
347 (Bouafif et al. 2008), and this makes the C3 and C4 more heterogeneous and complex.

348 The results show the reduction of 61% and 44% of carbon C1 in the pulps immersed in
349 OPC and geopolymer respectively, and the increase of 6% of C1 in the pulp immersed in CAC. In
350 the NFC, it is observed that the immersion of these fibers in OPC paste led to a reduction of 40%
351 of C1 and increase of 38% and 63% of C1 in the fibers immersed in CAC and geopolymer,
352 respectively. The reduction of carbon C1 indicates removal of lignin, or extractives, or both, from
353 the surface of the fibers. The increase in C1 indicates the degradation of hemicelluloses or
354 cellulose, leading to exposure of the lignin or extractives on the surface of the fibers. This
355 correlation is clear when considering the reduction of 11% and 18% of the carbon C2 on the
356 surface of NFC immersed in CAC and geopolymer, respectively; and the increase of 27% of C2 on
357 the surface of NFC in OPC. In the pulp, there was a reduction of 11% of C2 for immersion in CAC
358 and the increase of 6% of C2 in the pulp immersed in OPC.

359

360

361

Table 5 – Functional groups on the fibers surface

Fibers	Functional groups			
	% C1	% C2	% C3	% C4
	(C-C, C-H)	(C-O, C-OH)	(C=O, O-C-O)	(O-C=O)
Pulp control	8.09 ± 0.58	38.21 ± 0.91	45.18 ± 1.65	8.52 ± 0.88
Pulp OPC	3.05 ± 0.81	40.38 ± 1.01	46.87 ± 1.05	9.70 ± 0.51
Pulp CAC	8.60 ± 0.83	34.06 ± 1.89	48.36 ± 0.99	8.98 ± 1.10
Pulp geopolymer	4.52 ± 1.04	36.73 ± 1.11	47.97 ± 0.52	10.77 ± 1.26
NFC control	7.14 ± 0.97	34.60 ± 2.46	47.72 ± 1.77	10.54 ± 1.73
NFC OPC	4.19 ± 1.11	43.76 ± 1.60	42.59 ± 2.13	9.45 ± 0.47
NFC CAC	9.78 ± 0.94	30.85 ± 1.62	49.76 ± 1.49	9.60 ± 0.83
NFC geopolymer	11.65 ± 1.38	28.55 ± 1.44	48.70 ± 1.34	11.10 ± 0.55

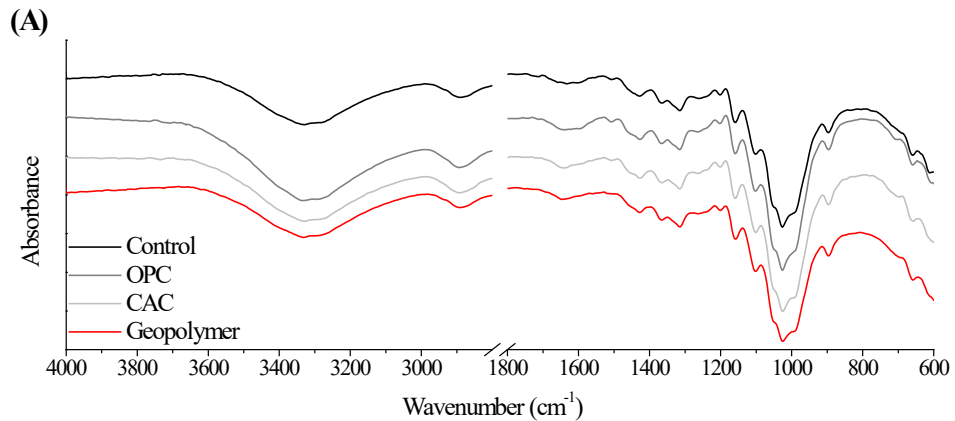
363

364 These results of the high-resolution XPS spectra confirm the low resistance of lignin,
365 especially in pastes based on OPC and geopolymer. This reinforces the assertion that the high
366 alkalinity dissolves the lignin and hemicelluloses of fiber, and besides the low durability of these
367 components in a highly alkaline environment, they are sensitive to Ca(OH)₂, present in the OPC
368 matrix. About the geopolymer, Ye et al. (2018) and Alshaaer et al. (2017) concluded that
369 geopolymerization causes alkaline degradation of lignin and hemicelluloses by the generation of
370 carboxylic acids.

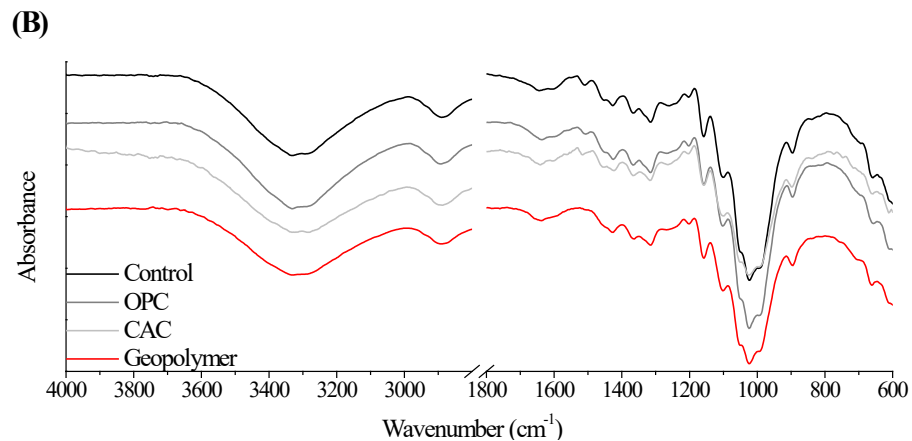
371 The degradation of the cellulose in the CAC and geopolymer pastes, indicated by the
372 reduction of the carbon C2 on the surface of the fibers, is according to Gram (1988) and Tolêdo
373 Filho et al. (2000). These authors have found that high alkalinity causes the hydrolysis of the
374 cellulose molecule, leading to degradation of molecular chains and then a reduction in the degree
375 of polymerization of the fibers.

376 The FTIR analysis of the pulp and NFC immersed in the different unhardened
377 cementitious pastes also indicates the modifications that occurred in the chemical structure of the
378 fibers due to the exposure in the alkaline environment. Hence, Figure 5 (A) and Figure 5 (B) show
379 the spectra of pulp and NFC respectively, in the control condition and after 28 days in the
380 unhardened pastes of the OPC, CAC and geopolymer.

381 The spectra display the characteristic peaks of cellulosic fibers, but it does not show clear
382 shifts in the bands due to the exposure of the fibers under different conditions. However, the
383 careful analysis of the OH stretching vibration peak intensities and positions within the range of
384 3000–3800 cm^{-1} , gives considerable information concerning the hydrogen bonds (Altaner et al.
385 2014; Dai and Fan 2011; Fengel 1992; 1993).



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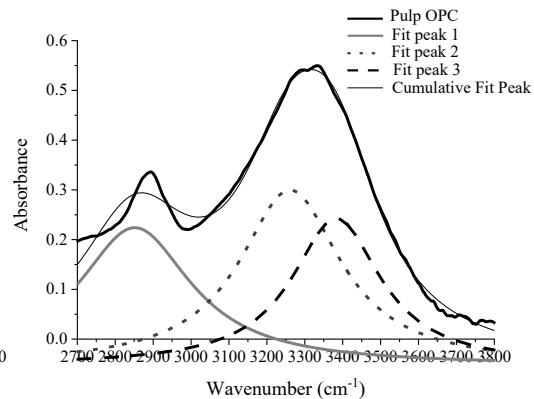
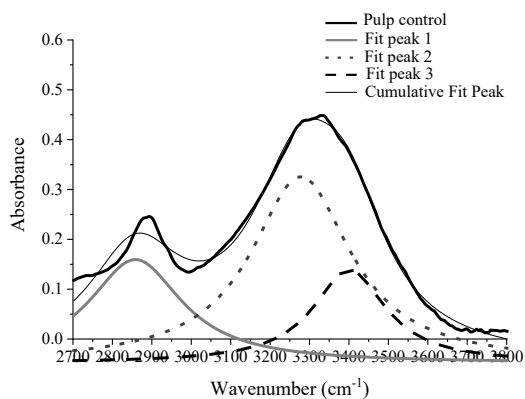
388 Figure 5 – FTIR spectra of pulp (A) and NFC (B) in the control condition and after 28 days of
389 immersion in the OPC, CAC and geopolymer pastes

390 * The IR region between 1800 cm^{-1} and 2700 cm^{-1} has been omitted from the spectra because it
391 does not contain any significant bands

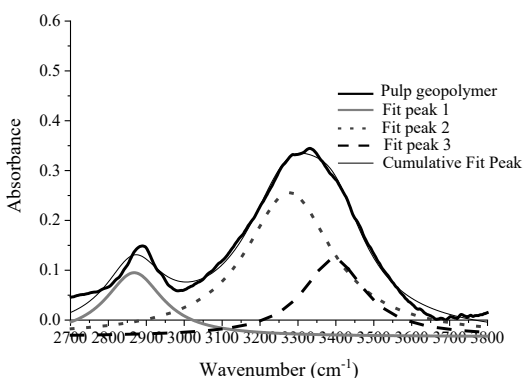
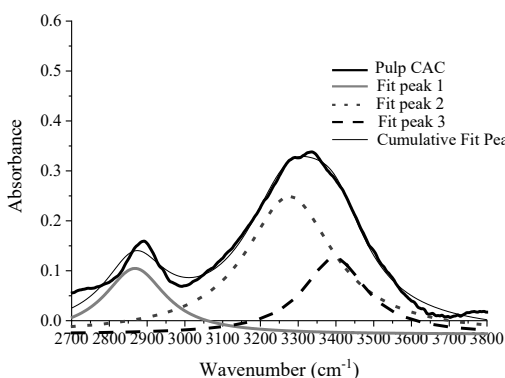
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393 Figure 6 and Figure 7 show the deconvoluted peaks into two bands for the curves fitting
394 of pulp and NFC in the control condition and immersed in different unhardened cement pastes.
395 The pulp and NFC control presented intermolecular hydrogen bonds, which are indicated in bands
396 at 3278 cm^{-1} and at 3269 cm^{-1} , and intramolecular hydrogen bonds at 3399 cm^{-1} and at 3394 cm^{-1} ,
397 respectively. According to Zhao et al. (2019) the intramolecular hydrogen bonds for O2H...O6
398 and O3H...O5, and the intermolecular hydrogen bonds for O6H...O3' in cellulose appear at 3455–

399 3410, 3375–3340, and 3310–3230 cm^{-1} , respectively, along with the valence vibration of H-
 400 bonded OH groups at 3570–3450 cm^{-1} .

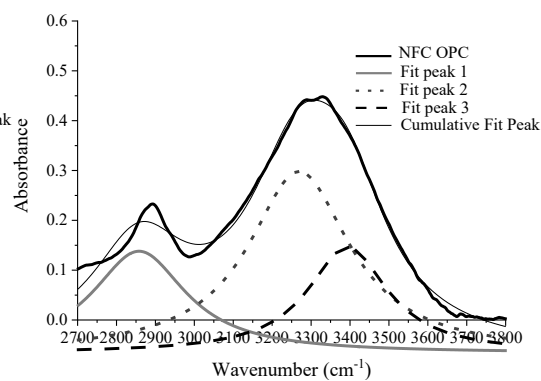
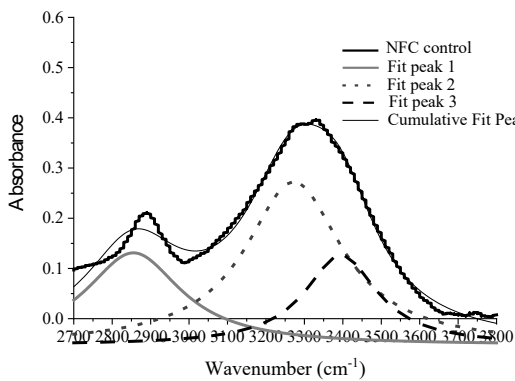


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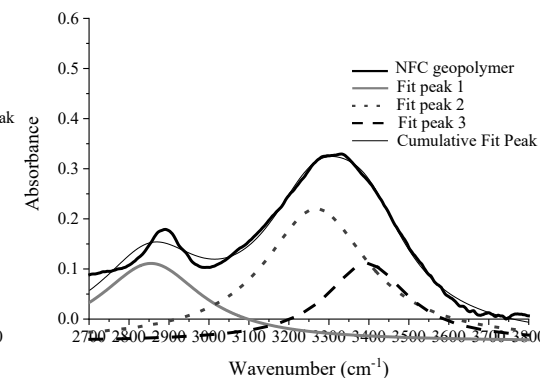
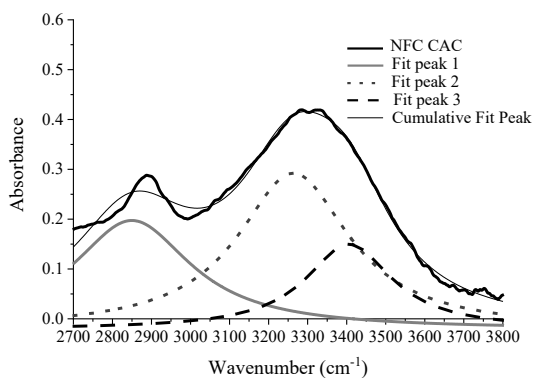


402

403 Figure 6 – Deconvoluted FTIR spectra of pulp control, and of pulp immersed in OPC, CAC and
 404 geopolymer pastes



405



406

407 Figure 7 – Deconvoluted FTIR spectra of NFC control, and NFC immersed in OPC, CAC and
 408 geopolymer pastes

409

410 It can be seen in Figure 6, Figure 7 and Table 6 that the immersion of the fibers in the
 411 alkaline environments (OPC, CAC and geopolymer) induced in general, the slight shift of the
 412 bands to lower wave numbers (red shift), compared to the control. The exception occurred in the
 413 bands related to intramolecular hydrogen bonds of NFC CAC and NFC geopolymer, which were
 414 shifted to higher wave numbers (blue shift).

415

416 Table 6. Band positions and its hydrogen bond energy

Fibers	Band position (cm^{-1})	E_H (kJ)	Band position (cm^{-1})	E_H (kJ)
Pulp control	3278.9 ± 1.3	26.6 ± 0.1	3399.0 ± 1.2	18.0 ± 0.2
Pulp OPC	3266.7 ± 1.2	27.4 ± 0.5	3386.6 ± 0.9	18.9 ± 0.4
Pulp CAC	3272.4 ± 0.6	27.1 ± 0.1	3393.5 ± 0.8	18.4 ± 0.2
Pulp geopolymer	3275.8 ± 1.2	26.9 ± 0.1	3394.4 ± 1.7	18.4 ± 0.1
NFC control	3269.2 ± 1.6	27.4 ± 0.1	3394.5 ± 1.2	18.4 ± 0.1
NFC OPC	3271.5 ± 1.7	27.2 ± 0.2	3393.2 ± 0.8	18.5 ± 0.1
NFC CAC	3263.9 ± 0.6	27.8 ± 0.0	3404.5 ± 1.8	17.6 ± 0.1
NFC geopolymer	3266.9 ± 2.2	27.5 ± 0.2	3395.4 ± 0.4	18.3 ± 0.0

417

418 The red shift of OH stretching frequency suggest the formation of the H bond and
 419 consequently, the weakening of the OH bond, causing elongation of these OH bonds (Hobza and
 420 Havlas, 2000 and Van der Veken et al., 2001). The formation of new hydrogen bonds can be
 421 confirmed with the results of the hydrogen bonding energy, which increased with the shift of the
 422 bands to lower wave numbers frequencies.

423 The shift of the bands of the NFC to higher wave numbers (NFC CAC (3404 cm^{-1}) and
 424 NFC geopolymer (3395 cm^{-1})), compared to the control, gives indications of the weakening of
 425 hydrogen bond and reduction of its energy. According to Turki et al. (2018) this may be explained
 426 by the formation of new OH bonds at this frequency, but these bonds are not as strong as the initial
 427 ones due to the lack of stability and to the packing effect that makes the length of OH covalent

428 bonds shorter. These modifications, blue and red shift, indicate changes in the chemical structure
429 and hydrogen bonds of the pulp and NFC after immersion in unhardened OPC, CAC and
430 geopolymer pastes. These changes can be associated with different patterns of geometrical
431 perturbations or electron density shifts (Scheiner and Kar 2002).

432 According to Ciolacu et al. (2011) the H bonding energy is directly related to the degree
433 of crystallinity of the cellulose. Thus, the increase in H bond energy indicates changes in the
434 crystalline structure of the cellulose, ie increase in the degree of crystallinity. Therefore, the
435 increase in H bond energy indicates the degradation of the amorphous components on the surface
436 of the fibers (pulp OPC, CAC, geopolymer, and NFC OPC) after immersion in alkaline pulps. On
437 the other hand, the reduction of H bond energy of NFC CAC and NFC geopolymer is related to the
438 reduction of crystallinity, which indicates degradation of the cellulose in these fibers.

439 The immersion of the pulp and NFC in CAC, OPC and geopolymer caused slight
440 modifications in the fibers, but that were sufficient to consider the alkaline environment of these
441 pastes as being harmful during the time of exposure of lignocellulosic fibers. This is because in
442 highly alkaline environments cellulose and hemicellulose are exposed to reactions such as alkaline
443 swelling; alkaline dissolution; alkaline hydrolysis of acetyl groups; peeling-off reaction; chemical
444 and physical stopping reaction and alkaline hydrolysis (Van Loon et al. 1999; Pavasars et al.
445 2003). According to the authors, the rates at which these reactions occur are dependent on pH,
446 temperature and alkaline environment.

447 According to Nevel (1985) and Askarieh et al. (2000) at temperatures below 170 °C the
448 predominant mechanism of alkaline cellulose degradation is by a reaction that takes place at the
449 reducing end group of the chain and ruptures the 1,4-glycosidic linkage. Glucose units are
450 progressively eliminated from the reducing end of the chain, and this process is known as the
451 peeling. Terminal degradation consists of shortening of the main chain by successive elimination
452 of the terminal reducing monomer. This reaction occurs in solutions with a pH greater than 11.
453 The cellulose molecule loses about 50 to 70 units of glucose in this type of hydrolysis (Grace et
454 al., 1989).

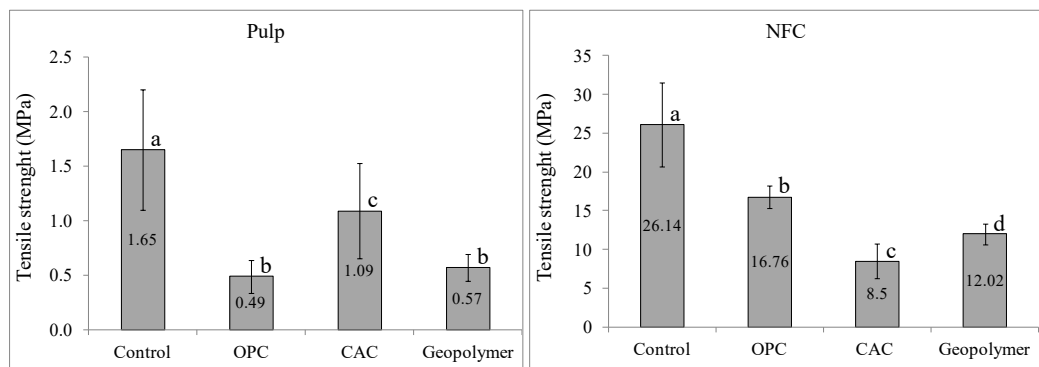
455 According to Johansson and Samuelson (1975) the peeling reaction rate is proportional to
456 the concentration of reducing end groups of the cellulose. Sixta (2006) confirm that the alkaline
457 solubilization occurs whenever any new region containing fibers becomes accessible or when there
458 is a significant reduction in fibers size, and that the low degree of polymerization (DP) of fibers

459 favors the alkaline solubilization phenomenon. This indicates that the intensity of alkaline
460 degradation is different in pulp and NFC.

461 The nanofibrillation process reduces considerably the DP of the fibers because the cell
462 wall structure consisting of nanofibers in a multilayered structure and hydrogen bonds is broken
463 down. According to Zimmermann et al. (2010) fibrillation in the laboratory led to a decrease in DP
464 between 15% and 63%. The DP correlates with the aspect ratio of the fibrils, and this means that,
465 shorter fibrils have a lower DP. According to Matsuoka et al. (2014) and Van Loon et al. (1999)
466 the large number of new reducing ends is formed in cellulose during the DP reduction, and the
467 extent of degradation depends on the number of reducing end groups, that in principle, would
468 enable complete degradation of cellulose. In other words, the greater the number of reducing end
469 groups, the higher the peeling reaction rate. In addition to the degree of polymerization, the higher
470 specific surface area of NFC, which greatly increases the quantity of surface hydroxyl groups,
471 increases the accessibility of this fiber to a greater extent of degradation.

472 The effect of the alkalinity of the OPC, CAC and geopolymer pastes on the mechanical
473 strength of the pulp sheets and the NFC is complementary to the study of the chemical degradation
474 on the surface of the fibers and can be verified in Figure 8.

475



476
477

*Average values followed by the same letters do not differ significantly by the Tukey test ($p < 0.05$).

478

Figure 8 – Tensile strength of pulp and NFC

479

480 The tensile strength results show that pulps and NFC immersed in the OPC, CAC and
481 geopolymer pastes had statistically significant reduction in the tensile strength, in comparison to
482 pulp and NFC control. However, the pulp immersed in CAC had the lower reduction rate (34%) in
483 the mechanical strength after 28 days of immersion than OPC and geopolymer (approximately
484 70%). For the NFC the effect of the matrix was different than that found for the pulp, where the

485 lower reduction rate in the mechanical strength was to OPC (36%). The immersion of NFC in
486 CAC and geopolymer caused the higher reductions in tensile strength, which were of 68% and
487 54%, respectively.

488 As previously reported, the rate of alkaline degradation that occurs in lignocellulosic
489 fibers is directly related to pH, temperature and alkaline environment. Table 3 shows that OPC has
490 higher pH than CAC and geopolymer. However, the CAC and geopolymer pastes released more
491 heat during hydration and geopolymerization reaction, as shown in Figure 2. Thus, NFC tensile
492 strength results show that the effect of heat release on CAC hydration and geopolymer reaction
493 was more prevalent in fiber degradation than the higher pH of OPC paste. This finding can be
494 confirmed by the reduction in glycosidic bonds (C-O-C) of pulps and NFCs immersed in CAC and
495 geopolymer, which is observed by the reduction of the C2 component (Table 5).

496 Thus, this study certified that immersion of cellulosic pulp and nanofibrillated cellulose in
497 the highly alkaline and unhardened cement paste exposes them to deterioration through the
498 chemical modifications of the extractives, residual lignin and carbohydrates (cellulose and
499 hemicellulose) in the surface of the fibers.

500

501 **Conclusions**

502

503 This study showed that not only Portland cement should be considered as harmful to
504 lignocellulosic fibers, through the migration of calcium hydroxide to the lumen and cell wall of the
505 fibers. Although CAC and geopolymer are free of calcium hydroxide, the high alkalinity of these
506 matrices also accelerated the degradation process of lignocellulosic fibers.

507 The surface components of unbleached bamboo pulp and NFC, such as lignin, extractives,
508 hemicellulose and cellulose, were modified after 28 days of immersion in the alkaline cement
509 pastes. The high alkalinity of the pastes also modified the hydrogen bonds in the fibers. These
510 changes in the chemical composition and molecular structure of the fibers directly affected the
511 strength of the pulp and nanofibrillated cellulose, confirming the degradation of the fibers.

512 This study also showed that the response of the bamboo pulp and NFC immersed in the
513 high alkaline cement pastes is different due the distinctions in characteristics of these fibers. The
514 combination of factors such as, the lower degree of polymerization and high specific surface area

515 of NFC with the highest amount of heat released during hydration of CAC and geopolymerization
516 favored the greatest degradation of NFC. The higher pH of OPC and the higher heat released in the
517 first hours of geopolymer reaction may have caused the more significant changes in the chemical
518 structure of the pulp and the reduction of its tensile strength.

519 Therefore, the best known means to reduce the degradation of the fibers are still those
520 which concomitant reduce the pH and calcium hydroxide of the matrices.

521

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523

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