

TESTING SYSTEMS TO MONITOR ATMOSPHERIC CO₂, PH AND PCO₂ IN THE OBSEA SUBMARINE OBSERVATORY

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Abstract— Due to the increase, in last years, of carbon dioxide (CO₂) level in the atmosphere we start monitoring with the expandable seafloor observatory OBSEA a variety of parameters related with ocean acidification. Our research team has installed an underwater pH sensor in the OBSEA observatory in order to compare the sea pH data with the CO₂ data collected with an atmospheric CO₂ sensor.

Keywords— acidification, pH, CO₂, pCO₂

I. INTRODUCTION

It is now well recognised that levels of carbon dioxide (CO₂) in the atmosphere have been increasing since the Industrial Revolution due to burning of fossil fuels, leading to global warming [1]. In fact, the level of CO₂ in Earth's atmosphere already passed the critical threshold of 400 parts per million in Mauna Loa, Hawaii on May, 2013 (<http://www.esrl.noaa.gov/gmd/ccgg/trends/weekly.html>). This is the highest concentration of CO₂ that the open atmosphere has sustained in probably around three million years [2] and is a clear signal of the negative impact of humans on the environment and the climate system.

Only part of this anthropogenic CO₂ is accumulated in the atmosphere (45%), the rest is taken up by oceans and land in similar parts [3]. Importantly, the oceanic counterpart (during the 2003-2012 decade, 2.9 gigatons of C dissolved in the oceans every year [4]) involves a progressive acidification of seawater, since once dissolved, CO₂ intervenes in a series of chemical reactions leading to a lowering in pH [5]. Today, global oceans have already acidified by 0.1 pH units since preindustrial times. Predictions indicate acidification in the order of 0.3 to 0.5 pH units by 2100 and of nearly 0.8 pH units by 2300, a scenario with no obvious precedents over the last tens of millions of years [5]. Such pH reduction could have major effects on marine biota, especially calcifying organisms including coral reef communities, which will be unable to calcify effectively under these new conditions (e.g. [6,7]).

The Mediterranean Sea has certain characteristics that make it especially sensitive to the anthropogenic increase in atmospheric CO₂. Owing to the higher salinity of its waters, they have higher alkalinity as well [8], and thus a greater chemical capacity to neutralize acid and take up anthropogenic CO₂. On the other hand, the shorter residence time of deep waters [9] implies a rapid penetration of anthropogenic CO₂ in this basin.

There is debate, however, on whether the acidification due to the marine absorption of anthropogenic CO₂ is taking place more rapidly in the Mediterranean than in the global oceans. In 2011, a first study suggested that the Mediterranean Sea was acidifying faster [10]. However, a more recent study still under review, indicates that even though the penetration of anthropogenic CO₂ is occurring faster in the Mediterranean, this does not translate into higher rates of acidification [11].

Compared to open seas, the confluence of multiple processes that takes place in coastal zones makes even harder to predict the future trends of seawater pH and the associated vulnerability in these areas. In coastal zones, acidification can also be caused by other processes including the microbial degradation of organic matter in areas with high nutrient loading, the introduction of acidic river water, and the upwelling of CO₂ enriched deep water ([12] and references therein), amongst others.

In this context, and in order to gain insight on the acidification problem in coastal areas of the Mediterranean Sea, we found important to initiate instrumental time-series of pH and other parameters of the CO₂ system in seawater at fixed stations of the Catalan Coast. Taking advantage of the unique expandable seafloor observatory OBSEA, located at 4 km off the Vilanova i la Geltru coast (40 km south of Barcelona), at a depth of 20 m in a fishing protected area [13], and as a joint collaboration between the SARTI Research Group (UPC), Institut de Ciències del Mar, CSIC, Monterey Bay Aquarium Research Institute and Institut Català de Ciències del Clima, we are testing systems in OBSEA to start monitoring a variety of parameters related with ocean acidification.

II. TESTBED SYSTEMS

We started by implementing a system to monitor atmospheric CO₂ in order to compare it with dissolved CO₂ in seawater and to assess potential air-sea fluxes of this greenhouse gas [14]. Ideally, this system should have been installed in the fixed surface buoy that marks the OBSEA position. However, to facilitate maintenance and keep a better control of the system, we installed it in the shore station in Vilanova i la Geltrú, with an air inlet in the roof, and a 15 m Bev-A-Line tube line to the CO₂ analyser, a Licor 820 (Fig 1). In combination with data on wind direction, for comparisons with seawater pCO₂, we will only use the atmospheric CO₂ values during the periods in which the winds come directly from the sea.



Fig 1. Atmospheric CO₂ measurement system

In addition, we are also testing a relatively new system to measure pH (Fig 2), which uses ion sensitive field effect transistor (ISFET) technology [15,16]. This system has successfully been used in the past by several research groups (e.g. [17,18]). We are comparing these measurements with those from another autonomous pH sensor, the SAMI-pH from Sunburst Sensors, and with measurements performed in the laboratory using spectrophotometry [19]. So far the system seems to accurately measure pH, being sensitive to abrupt pH lowering events associated to storms and strong seawater mixing.

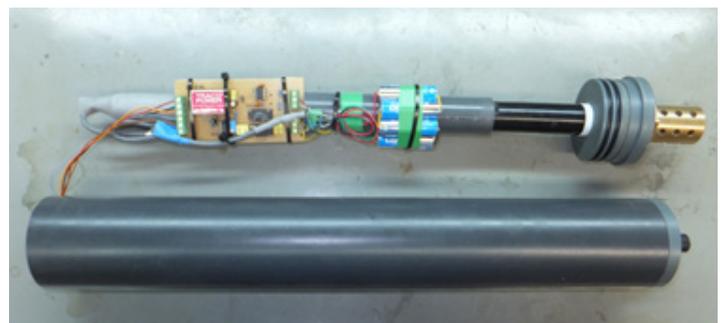


Fig 2. pH sensor based on ISFET technology

Finally, we are also attempting tests with a pCO₂ autonomous sensor (Fig 3), a CO₂-Pro from Pro-Oceanus [20]. Data from this system, in combination with atmospheric CO₂, should allow an assessment of air-sea CO₂ fluxes, to evaluate the role of this coastal area as a sink or source of atmospheric CO₂, a role that probably changes seasonally.



Fig 3. Pro-Oceanus CO₂-Pro CO₂ sensor

We hope we will be able to have this entire systems operative at the OBSEA cabled observatory in the near future, providing real-time observations of these parameters, perhaps with the possible addition of complementary sensors to allow a more complete determination of all seawater CO₂ system parameters. In this sense, the autonomous measurement of seawater alkalinity seems now a real possibility, after the very recent development of the so called SAMI-alk by Sunburst Sensors [21].

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REFERENCES

[1] Le Quééré, C., Raupach, M. R., Canadell, J. G. & Marland et al., G. Trends in the sources and sinks of carbon dioxide. *Nature Geoscience* 2, 831 – 836 (2009).
 [2] Pagani, M., Liu, Z., LaRiviere, J. & Ravelo, A. C. High Earth-system climate sensitivity determined from Pliocene carbon dioxide concentrations. *Nature Geosci* 3, 27-30, (2010).
 [3] Ballantyne, A. P., Alden, C. B., Miller, J. B., Tans, P. P. & White, J. W. C. Increase in observed net carbon dioxide uptake by land and oceans during the past 50 years. *Nature* 488, 70-72 (2012).

[4] Le Quééré, C. et al. Global carbon budget 2013. *Earth Syst. Sci. Data* 6, 235-263, (2014).
 [5] Pelejero, C., Calvo, E. & Hoegh-Guldberg, O. Paleo-perspectives on ocean acidification. *Trends in Ecology & Evolution* 25, 332-344 (2010).
 [6] Movilla, J. et al. Calcification reduction and recovery in native and non-native Mediterranean corals in response to ocean acidification. *J. Exp. Mar. Biol. Ecol.* 438, 144-153 (2012).
 [7] Bramanti, L. et al. Detrimental effects of ocean acidification on the economically important Mediterranean red coral (*Corallium rubrum*). *Global Change Biology* 19, 1897–1908, (2013).
 [8] Schneider, A., Wallace, D. W. R. & Kortzinger, A. Alkalinity of the Mediterranean Sea. *Geophys. Res. Lett.* 34, L15608, doi:15610.11029/12006GL028842 (2007).
 [9] Bethoux, J. P., El Boukhary, M. S., Ruiz-Pino, D., Morin, P. & Copin-Montegut, C. in *The Mediterranean Sea Vol. 5K The handbook of environmental chemistry* (ed A. Salot) 67-86 (Springer, 2005).
 [10] Touratier, F. & Goyet, C. Impact of the Eastern Mediterranean Transient on the distribution of anthropogenic CO₂ and first estimate of acidification for the Mediterranean Sea. *Deep Sea Research Part I: Oceanographic Research Papers* 58, 1-15 (2011).
 [11] Palmiéri, J. et al. Simulated anthropogenic CO₂ uptake and acidification of the Mediterranean Sea. *Biogeosciences Discuss.* 11, 6461-6517, (2014).
 [12] Wallace, R. B., Baumann, H., Grear, J. S., Aller, R. C. & Gobler, C. J. Coastal ocean acidification: The other eutrophication problem. *Estuar. Coast. Shelf S.* 148, 1-13, (2014).
 [13] Mánuel-Lázaro, A., Nogueras, M. & Del Rio, J. OBSEA: An Expandable Seafloor Observatory. *Sea Technology* 51, 37-39 (2010).
 [14] Wanninkhof, R. Relationship between wind speed and gas exchange over the ocean revisited. *Limnol. Oceanogr. Meth.* 12, 351-362, (2014).
 [15] Martz, T. R., Connery, J. G. & Johnson, K. S. Testing the Honeywell Durafet (R) for seawater pH applications. *Limnology & Oceanography Methods* 8, 172-184 (2010).
 [16] Artero, C., Nogueras, M. & Mánuel, A. pH Sensor. *Instrumental Viewpoint* 13, 23-24 (2012).
 [17] Price, N. N., Martz, T. R., Brainard, R. E. & Smith, J. E. Diel variability in seawater pH relates to calcification and benthic community structure on coral reefs. *PLoS ONE* 7, e43843 (2012).
 [18] Hofmann, G. E., Blanchette, C. A., Rives, E. B. & Kapsenberg, L. Taking the pulse of marine ecosystems: The importance of coupling long-term physical and biological observations in the context of global change biology. *Oceanography* 26, 140–148 (2013).
 [19] Clayton, T. D. & Byrne, R. H. Spectrophotometric seawater pH measurements: total hydrogen ion concentration scale calibration of m-cresol purple and at-sea results. *Deep-Sea Res.* 140, 2115-2129 (1993).
 [20] Jiang, Z. P. et al. Application and assessment of a membrane-based pCO₂ sensor under field and laboratory conditions. *Limnol. Oceanogr. Meth.* 12, 264-280, (2014).
 [21] Spaulding, R. S. et al. Autonomous in Situ Measurements of Seawater Alkalinity. *Environmental Science & Technology* 48, 9573-9581, (2014).